

Theoretical Studies on Recent Discovered Organic Superconductors

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Theoretical Studies on Recent Discovered Organic Superconductors

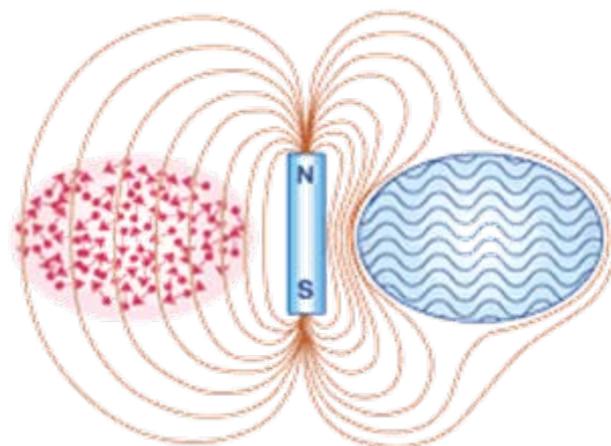
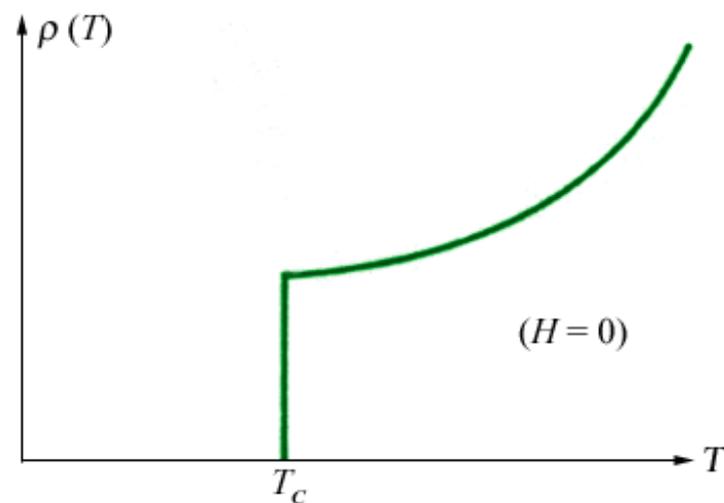
Abstract: In this talk, I report our recent study on the magnetic and pair binding properties in newly discovered polycyclic aromatic hydrocarbon (PAH) superconductors including alkali-metal-doped picene, coronene, phenanthrene, and dibenzopentacene. To gain a better understanding on magnetism and electron correlation in PAH, we have performed a systematic numerical investigation on the correlation effects and model the π -electrons on the carbon atoms of a single molecule by a one-orbital Hubbard model, in which the energy difference ϵ between carbon atoms with and without hydrogen bonds is taking into account. We demonstrate that the spin polarized ground state is realized for charged molecules in the physical parameter regions, which provides a reasonable explanation of local spins observed in PAHs. In alkali-metal-doped dibenzopentacene, our results show that electron correlation may produce an effective attraction between electrons for the charged molecule with one or three added electrons. We also propose a different doping pattern which may lead to higher transition temperature. Some results on the possible structure of PAH as functions of pressure and doping will be discussed.

Outline

- Introduction: superconductivity and recent discoveries
- Possible structures
- Correlation effect
- Magnetic and Superconducting Instabilities
- Co-doped phenanthrene, new superconductor?
- Summary and discussions

What is a Superconductor?

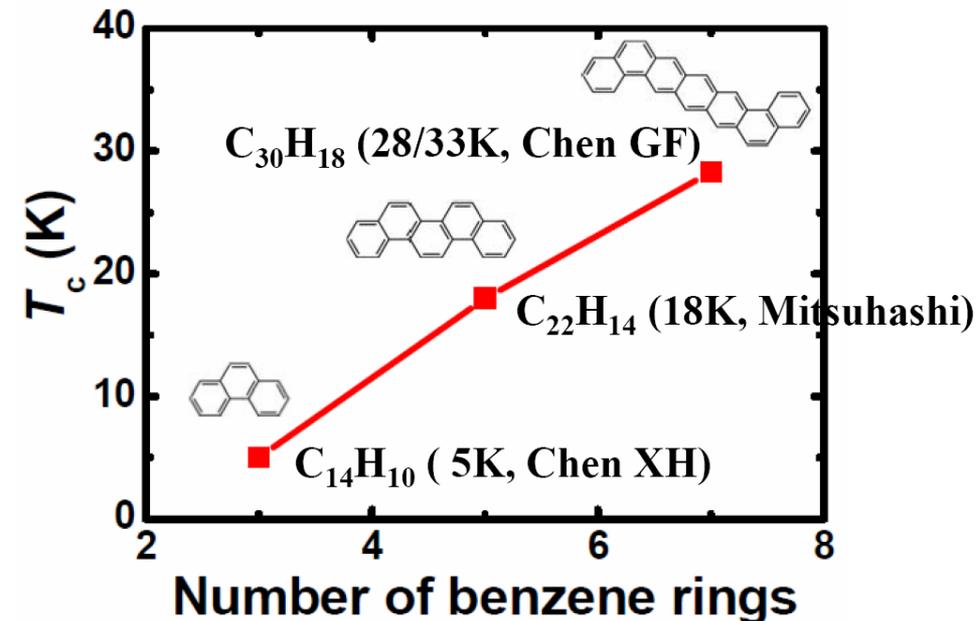
- Zero resistant
Perfect conductor.
 $\rho = 0$ for $T < T_c$.
No energy loss.
- Zero Magnetic Flux:
The Meissner Effect
Perfect Diamagnet.
- Critical Temperature
 T_c is the key factor.



Existing Superconductors

- **Metallic:** normal metals when $T > T_c$.
Conventional, low $T_c \leq 23.3\text{K}$ (39K).
- **Organic:** semiconductors when $T > T_c$.
New, low $T_c \leq 17\text{K}$ (before 2005); **2011, 33K?**
- **Ceramic:** complicated behavior when $T > T_c$.
Novel, high $T_c \approx 165\text{K}$, under pressure.
- **Fullerenes:** insulating/semiconducting when $T > T_c$.
Novel, intermediate $T_c \approx 35\text{K}$.
- **Mg-B:** metallic, $T_c \approx 39\text{K}$
- **Fe-based:** more complicated behavior, high $T_c \approx 55\text{K}$
- Others?
Room temperature?

Organic Superconductors



discoveries

Polycyclic Aromatic Hydrocarbon (PAH)

Picene (5-rings, $C_{22}H_{14}$) [1]; (coronene [2]);

Phenanthrene (3-rings, $C_{14}H_{10}$) [3, 4];

Dibenzopentacene (7-rings, $C_{30}H_{18}$) [5].

Superconducting: when doped by alkali-metal

Picene ($C_{22}H_{14}$) (2009-03-14, 2010-01-20)

nature

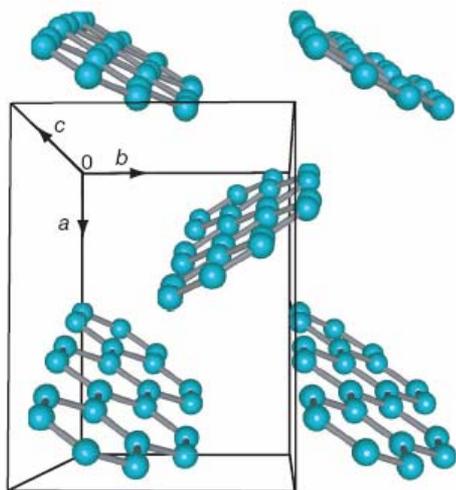
Vol 464 | 4 March 2010 | doi:10.1038/nature08859

LETTERS

First organic hydrocarbon superconductor

Superconductivity in alkali-metal-doped picene

Ryoji Mitsuhashi¹, Yuta Suzuki², Yusuke Yamanari², Hiroki Mitamura¹, Takashi Kambe², Naoshi Ikeda², Hideki Okamoto^{3,4}, Akihiko Fujiwara⁵, Minoru Yamaji⁶, Naoko Kawasaki¹, Yutaka Maniwa⁷ & Yoshihiro Kubozono¹



$T_C \sim 7\text{--}18\text{K}$

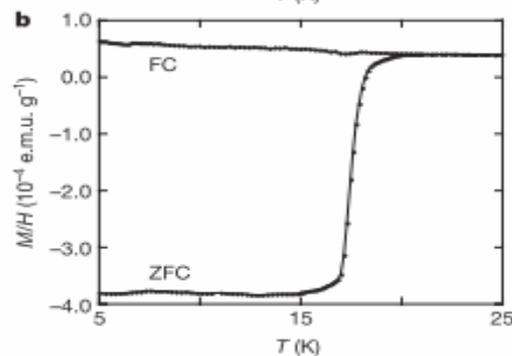
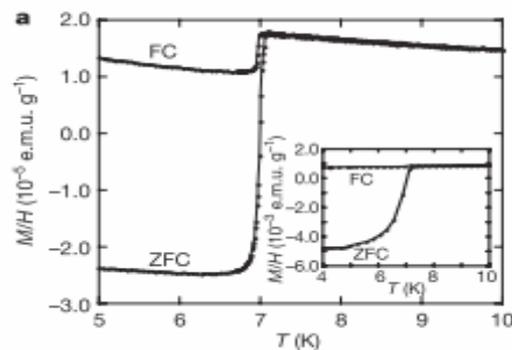
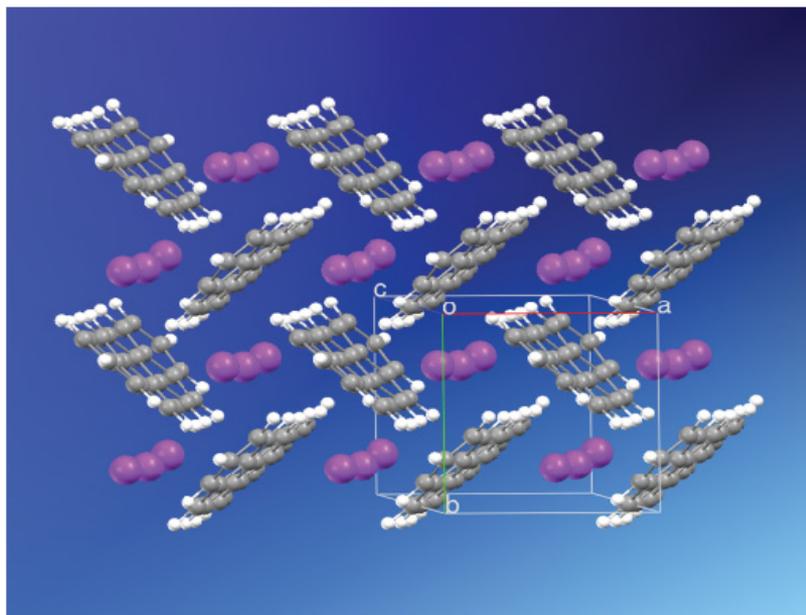
Table 1 | List of A_x picene (A: alkali-metal) samples prepared in this study

A	x	Annealing temperature (K)	Annealing time (days)	Physical properties	Shielding fraction
K	1.0	440	6.5	Pauli-like	NA
K	1.8	440	7.0	Pauli-like	NA
K	2.6	440	8.0	SC ($T_c = 6.5\text{ K}$)	$\ll 0.1\%$
K \dagger	2.9	440	9.0	SC ($T_c = 7.0\text{ K}$)	0.1%
K	3.0	440	8.0	SC ($T_c = 6.5\text{ K}$)	$\ll 0.1\%$
K	3.0	440	9.0	SC ($T_c = 17\text{ K}$)	0.1%
K	3.1	440	4.0	SC ($T_c = 7.4\text{ K}$)	$< 0.1\%$
K	3.3	440	21.0	SC ($T_c = 8\text{ K}$)	$\ll 0.1\%$
*K \ddagger	3.3	440	21.0	SC ($T_c = 6.9\text{ K}$)	15%
K	3.3	440	8.5	SC ($T_c = 7.1\text{ K}$)	$\ll 0.1\%$
K	3.3	440	11.0	SC ($T_c = 18\text{ K}$)	0.55%
*K \S	3.3	440	11.0	SC ($T_c = 18\text{ K}$)	1.2%
K	4.0	440	8.0	Curie-like	NA
K	5.1	440	12.5	Curie-like	NA
Na	3.4	570	5.0	Pauli-like	NA
Rb	2.8	440	16.5	Pauli-like	NA
RbI	3.1	570	6.7	SC ($T_c = 6.9\text{ K}$)	10%
Cs	3.0	440	9.0	Metal-insulator transition	NA

LETTERS

Superconductivity in alkali-metal-doped picene

Ryoji Mitsuhashi¹, Yuta Suzuki², Yusuke Yamanari², Hiroki Mitamura¹, Takashi Kambe², Naoshi Ikeda², Hideki Okamoto^{3,4}, Akihiko Fujiwara⁵, Minoru Yamaji⁶, Naoko Kawasaki¹, Yutaka Maniwa⁷ & Yoshihiro Kubozono¹



Phenanthrene ($C_{14}H_{10}$) (2011-07-13, 09-21)

NATURE COMMUNICATIONS | ARTICLE

Superconductivity at 5 K in alkali-metal-doped phenanthrene

X.F. Wang, R.H. Liu, Z. Gui, Y.L. Xie, Y.J. Yan, J.J. Ying, X.G. Luo & X.H. Chen

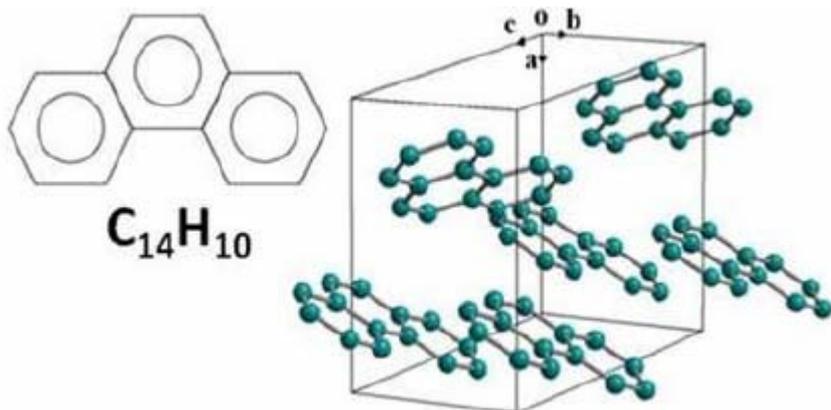
Affiliations | Contributions | Corresponding author

(Submitted on 20 Feb 2011)

arXiv:1102.4075v1

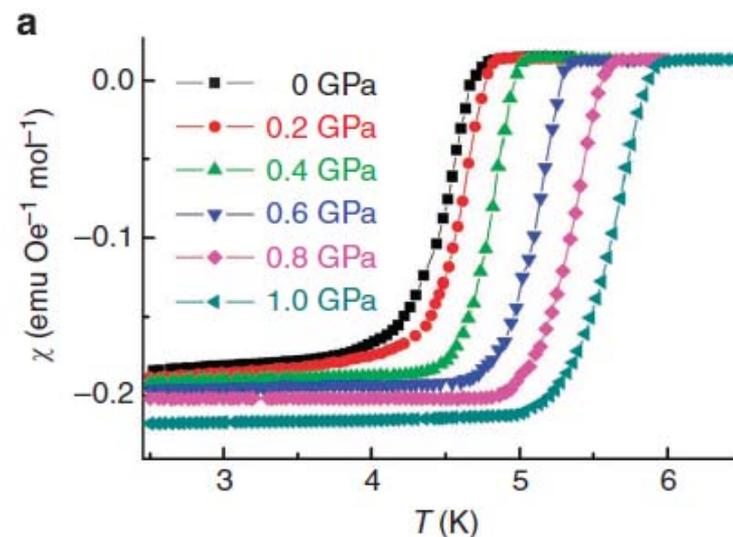
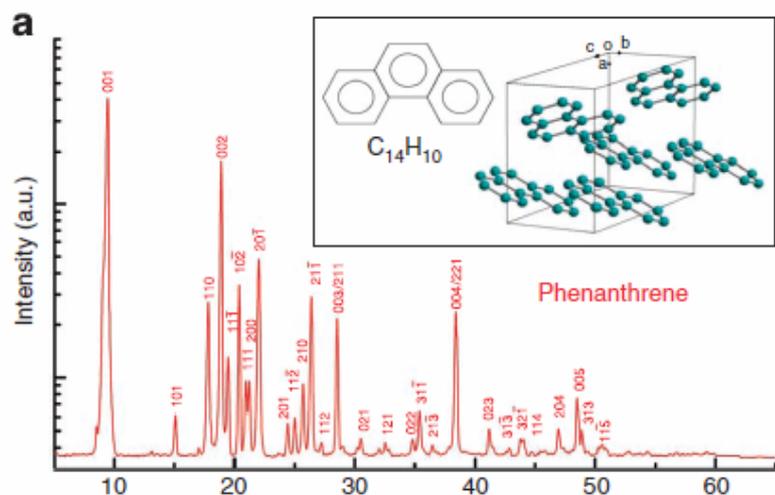
Nature Communications 2, Article number: 507 | doi:10.1038/ncom1507

Received 13 July 2011 | Accepted 21 September 2011



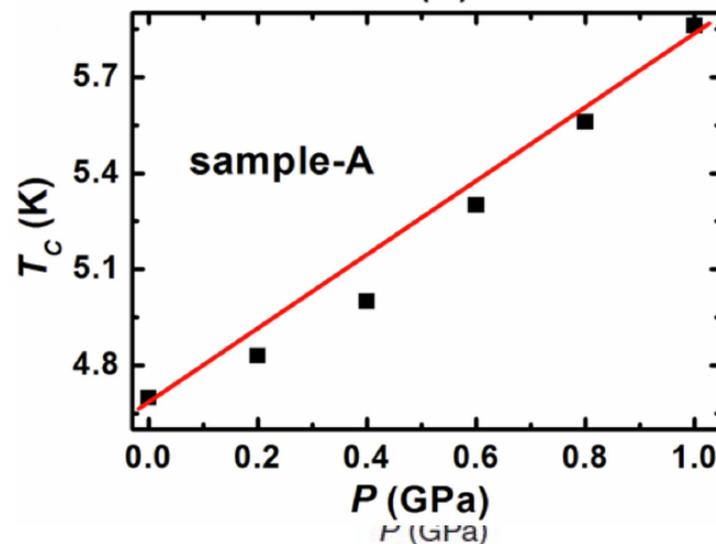
N.O.	Nominal composition	Annealing temperature	Annealing time	Onset SC transition	Shielding fraction
Sample A	K_3Ph	200°C	20 hours	4.7 K	1.9%
Sample B	K_3Ph	200°C	20 hours	4.9 K	1.1%
Sample C	K_3Ph	200°C	20 hours	4.7 K	4.3%
Sample D	$K_{2.5}Ph$	200°C	20 hours	NO SC	-
Sample E	$K_{2.8}Ph$	200°C	20 hours	NO SC	-
Sample F	$K_{2.9}Ph$	200°C	20 hours	NO SC	-
Sample G	$K_{3.1}Ph$	200°C	20 hours	NO SC	-
Sample H	$K_{3.2}Ph$	200°C	20 hours	NO SC	-
Sample I	$K_{3.5}Ph$	200°C	20 hours	NO SC	-

Phenanthrene ($C_{14}H_{10}$) (experiment)



$T_c \sim 5$ K

At 1 GPa, 20% increase of T_c





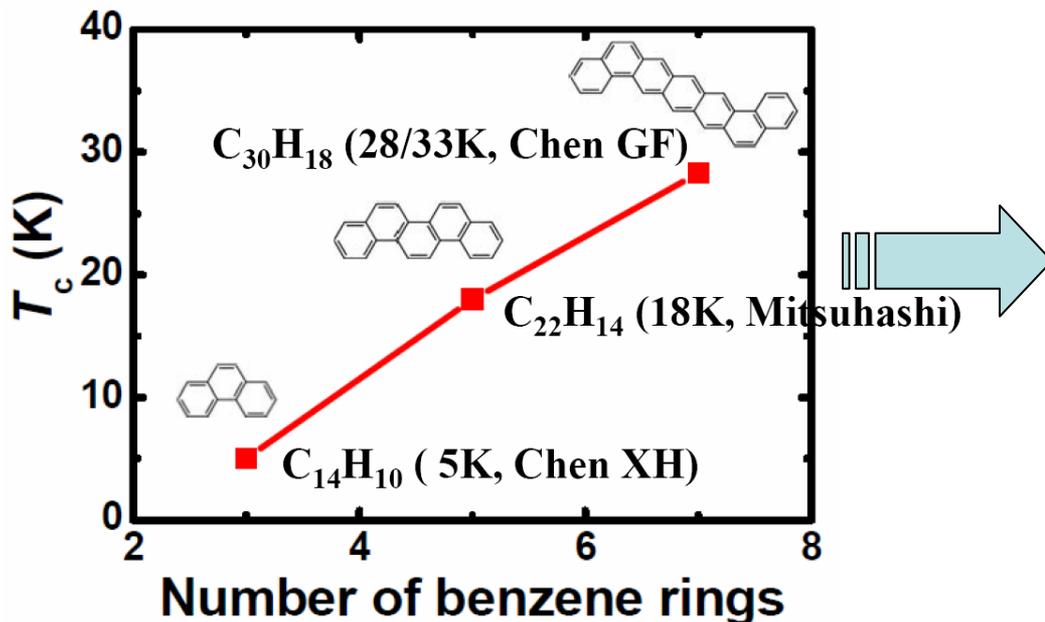
Superconductivity above 30 K in alkali-metal-doped hydrocarbon

Mianqi Xue^{1,2}, Tingbing Cao², Duming Wang³, Yue Wu¹, Huaixin Yang¹, Xiaoli Dong¹, Junbao He³, Fengwang Li² & G. F. Chen^{1,3}

¹Institute of Physics and Beijing National Laboratory for Condensed Matter Physics, Chinese Academy of Sciences, Beijing 100190, China, ²Department of Chemistry, Renmin University of China, Beijing 100872, China, ³Department of Physics, Renmin University of China, Beijing 100872, China.

SUBJECT AREAS:
 SUPERCONDUCTIVITY
 SUPERCONDUCTING MATERIALS
 MATERIALS PHYSICS
 CONDENSED MATTER PHYSICS

Organic Superconductors



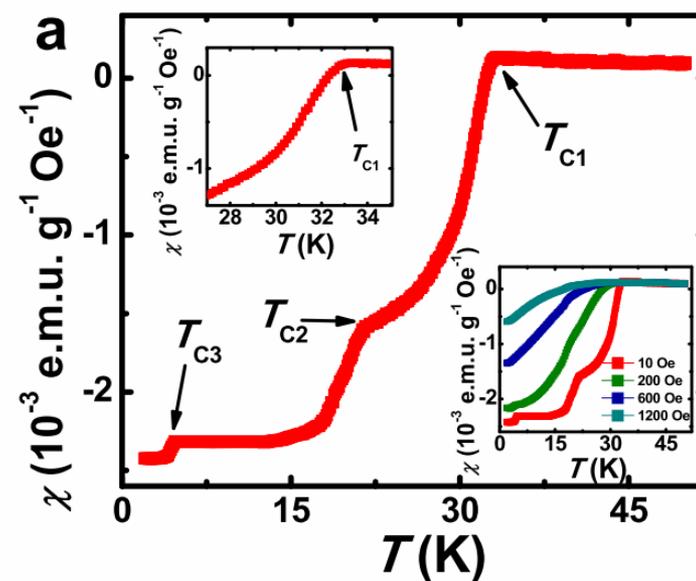
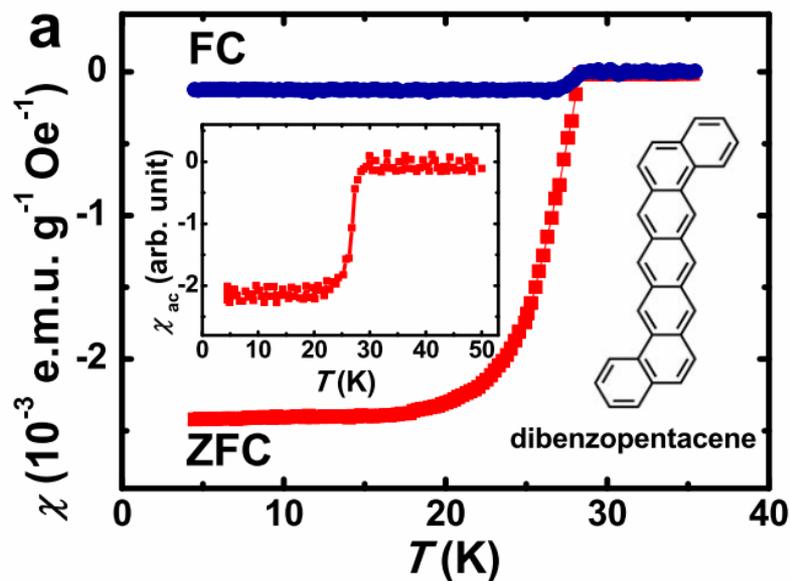
After Copper- and Iron-base superconductivity, a new superconductor family is discovered.

(Submitted on 3 Nov 2011)
[arXiv:1111.0820v1](https://arxiv.org/abs/1111.0820v1)

SCIENTIFIC REPORTS | 2 : 389 | DOI: 10.1038/srep00389

Superconductivity above 30 K in alkali-metal-doped hydrocarbon

Mianqi Xue^{1,2}, Tingbing Cao², Duming Wang³, Yue Wu¹, Huaixin Yang¹, Xiaoli Dong¹, Junbao He³, Fengwang Li² & G. F. Chen^{1,3}



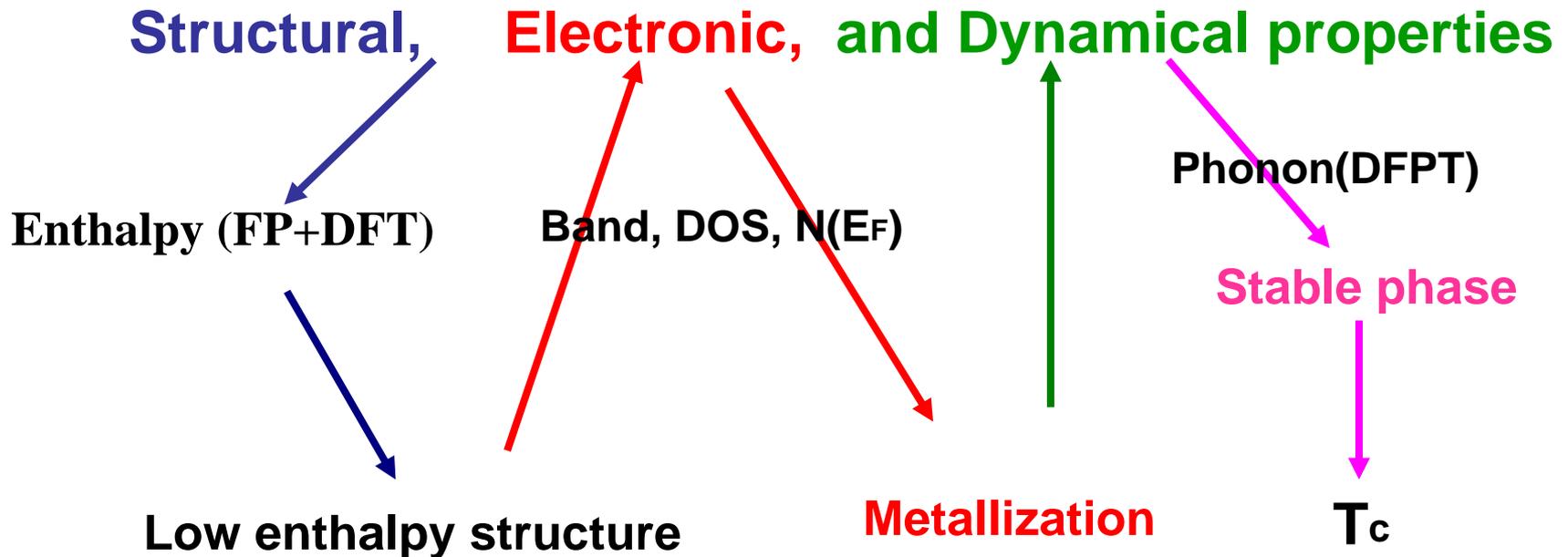
$T_c \sim 28-33$ K

Possible Structures

Searching for optimal structure

1. Undoped, $P = 1$ atm, most are known;
2. Doped $C_{22}H_{14}$, some studies;
3. Doped $C_{14}H_{10}$, some studies;
4. Doped $C_{30}H_{18}$ and others, no studies;
5. Few results as functions of pressure.

Computational Approach



- **Issues: Structure phase transition; Metallization pressure; Electronic and Dynamic properties; Superconductivity.**
- **Codes: CASTEP, Quantum Espresso, etc.**
- **When determining stable structure, zero point energy included.**

Picene ($C_{22}H_{14}$) Structure

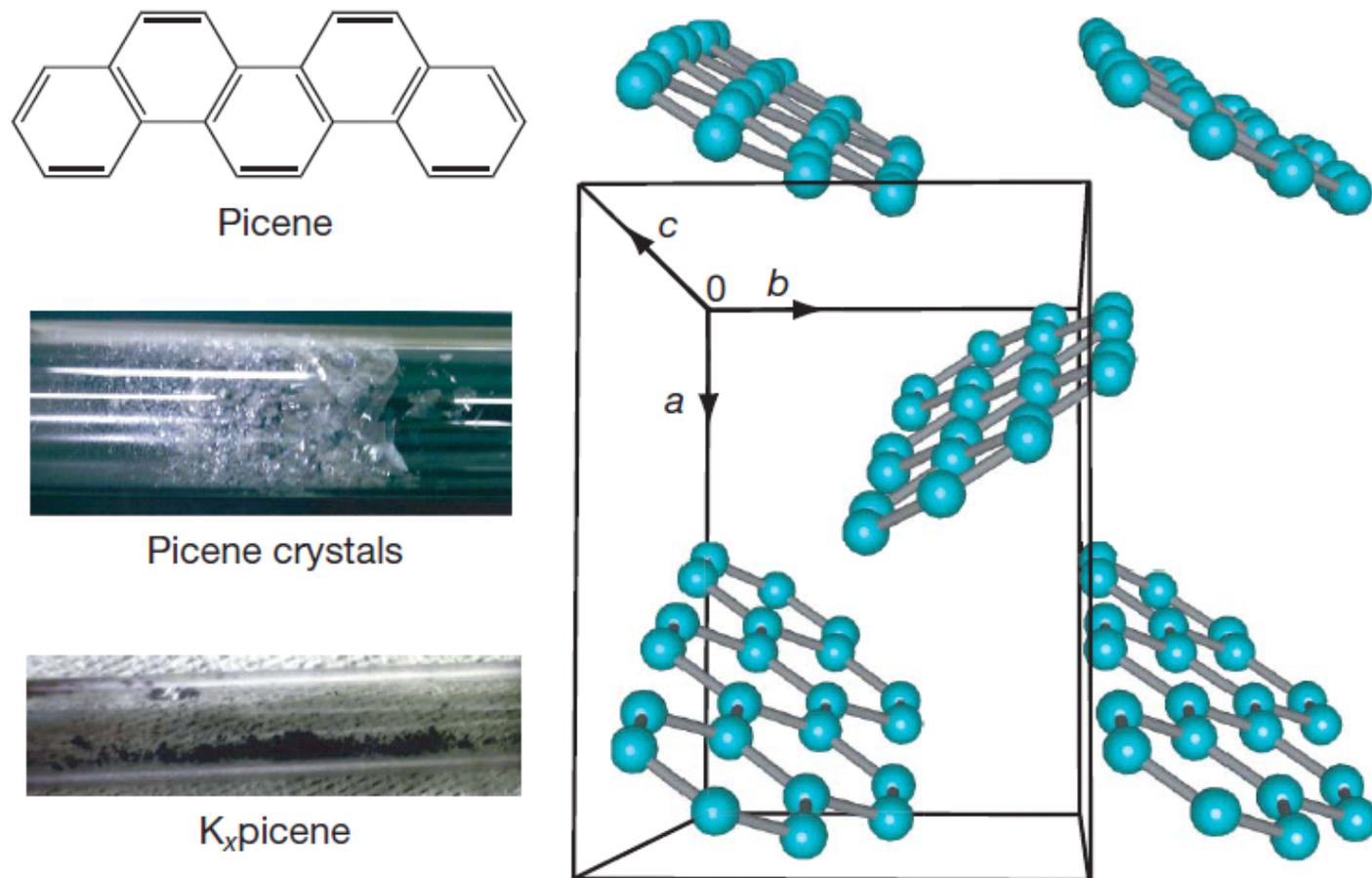


Figure 1 | Molecular structure, crystal structure and physical appearance of picene. Photographs show pristine picene (top; white) and K_xpicene (bottom; black).

$C_{22}H_{14}$ (theory)

First-Principles Electronic Structure of Solid Picene

Taichi KOSUGI^{1,2}, Takashi MIYAKE^{2,4}, Shoji ISHIBASHI²,
Ryotaro ARITA^{3,4}, and Hideo AOKI¹

Journal of the Physical Society of Japan
Vol. 78, No. 11, November, 2009, 113704
©2009 The Physical Society of Japan

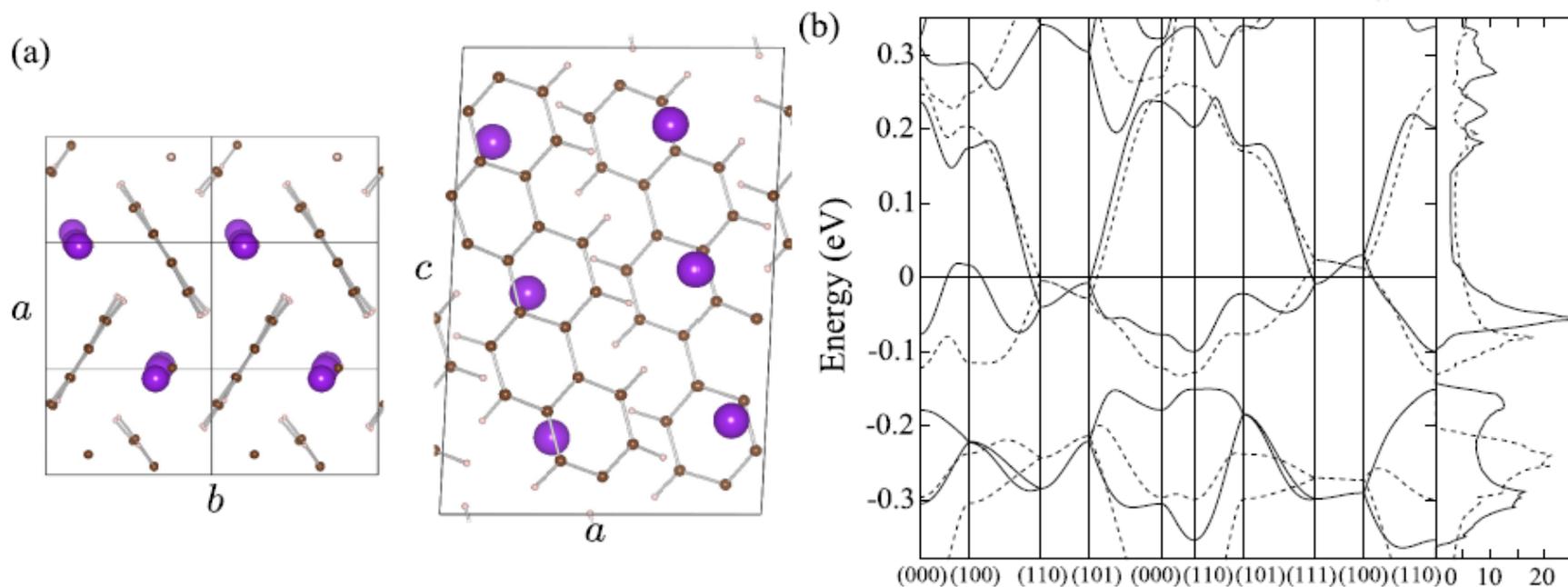
¹*Department of Physics, University of Tokyo, Bunkyo, Tokyo 113-0033*

²*Research Institute for Computational Sciences (RICS), AIST, Tsukuba, Ibaraki 305-8568*

³*Department of Applied Physics, University of Tokyo, Bunkyo, Tokyo 113-8656*

⁴*Japan Science and Technology Agency, CREST, Kawaguchi, Saitama 332-0012*

(Received August 14, 2009; accepted August 21, 2009; published November 10, 2009)



arXiv:1109.2059v1
(Submitted on 9 Sep 2011)

$C_{22}H_{14}$ (theory)

First-principles structural optimization and electronic structure of picene superconductor for various potassium-doping levels

Taichi Kosugi¹, Takashi Miyake^{1,2}, Shoji Ishibashi¹, Ryotaro Arita^{2,3,4}, and Hideo Aoki⁵

¹Nanosystem Research Institute "RICS", AIST, Umezono, Tsukuba 305-8568, Japan

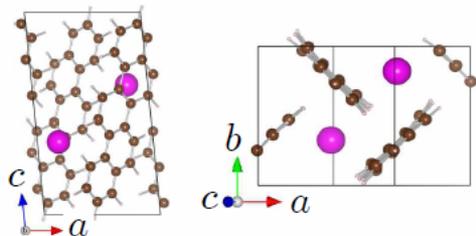
²Japan Science and Technology Agency (JST), CREST, Honcho, Kawaguchi, Saitama 332-0012, Japan

³Department of Applied Physics, University of Tokyo, Hongo, Tokyo 113-8656, Japan

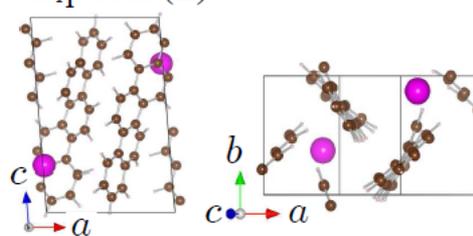
⁴Japan Science and Technology Agency (JST), PRESTO, Kawaguchi, Saitama 332-0012, Japan and

⁵Department of Physics, University of Tokyo, Hongo, Tokyo 113-0033, Japan

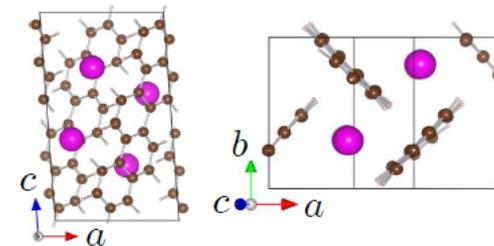
K_1 picene (A)



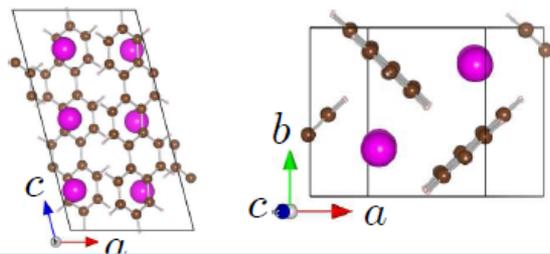
K_1 picene (B)



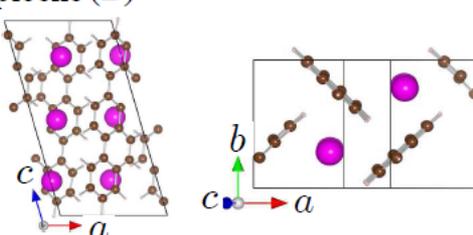
K_2 picene



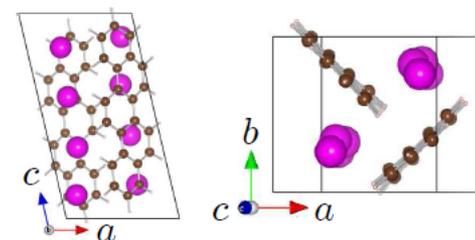
K_3 picene (A)



K_3 picene (B)



K_4 picene



Searching Optimal Structure is Undergoing

$C_{14}H_{10}$ (theory)

PHYSICAL REVIEW B 84, 144501 (2011)

Ab initio electronic and geometrical structures of tripotassium-intercalated phenanthrene

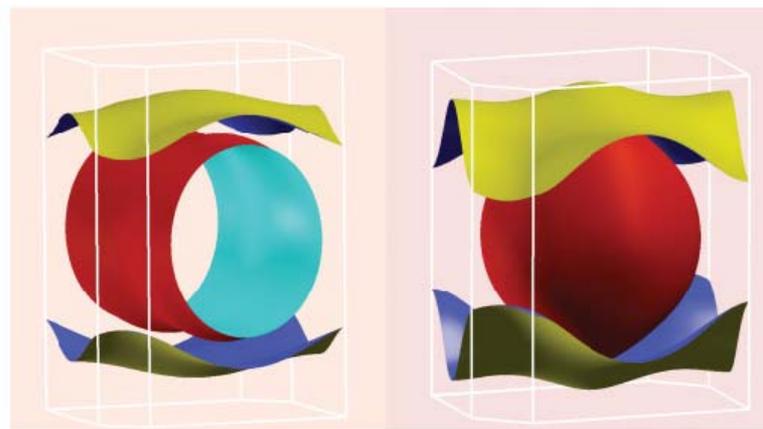
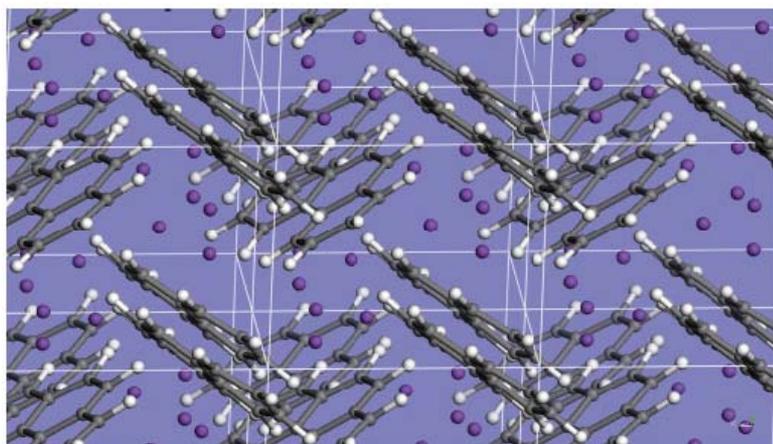
P. L. de Andres,^{1,*} A. Guijarro,² and J. A. Vergés³

¹*Donostia International Physics Center (DIPC), Paseo Manuel Lardizabal 4, E-20018 San Sebastian, Spain*

²*Departamento de Quimica Organica and Instituto Universitario de Sintesis Organica, Universidad de Alicante, San Vicente del Raspeig, E-03690 Alicante, Spain*

³*Departamento de Teoria y Simulacion de Materiales, Instituto de Ciencia de Materiales de Madrid (CSIC), Cantoblanco, E-28049 Madrid, Spain*

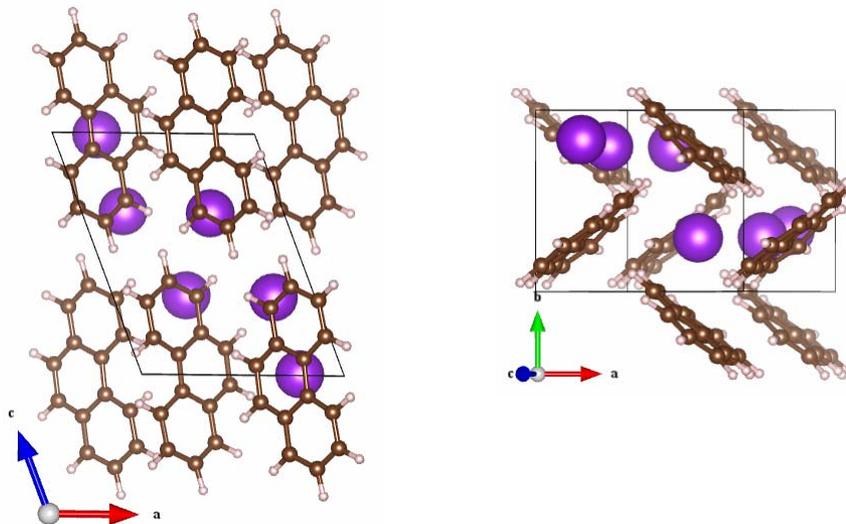
(Received 26 July 2011; published 3 October 2011)



Two sheets Fermi surface -> 1D & 2D Characters

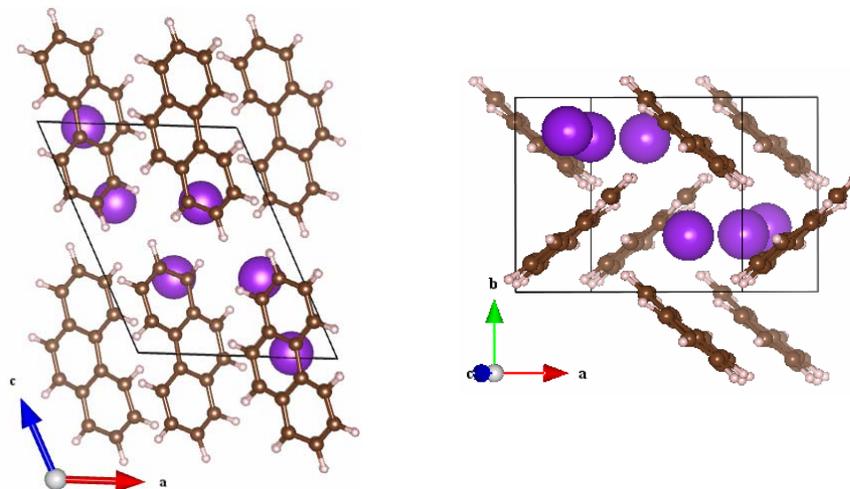
$\text{K}_3\text{C}_{14}\text{H}_{10}$ (our results)

LDA



$$\begin{aligned} a &= 7.758 \text{ \AA}, \\ b &= 6.852 \text{ \AA}, \\ c &= 10.039 \text{ \AA}; \\ \varrho &= 110.6^\circ, \\ V &= 499.36 \text{ \AA}^3 \end{aligned}$$

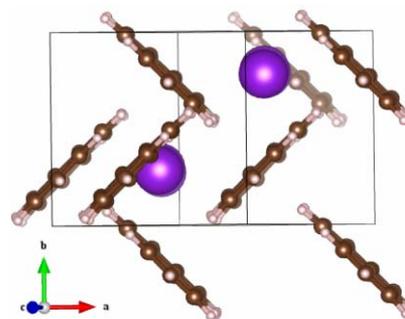
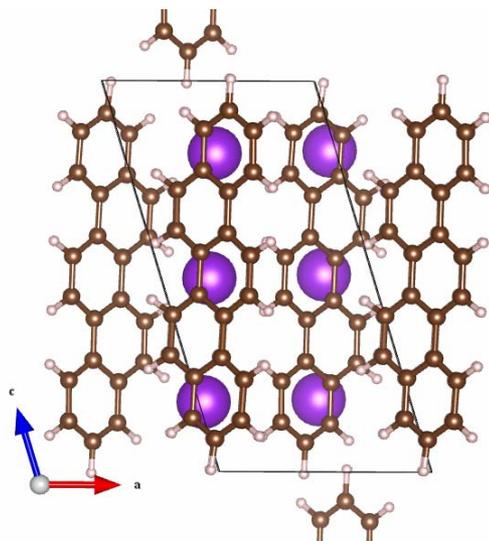
GGA-PBE



$$\begin{aligned} a &= 8.358 \text{ \AA}, \\ b &= 7.169 \text{ \AA}, \\ c &= 10.687 \text{ \AA}; \\ \varrho &= 115.1^\circ, \\ V &= 580.13 \text{ \AA}^3 \end{aligned}$$

$\text{K}_3\text{C}_{22}\text{H}_{14}$ (our results)

LDA



$$a=7.316\text{\AA},$$

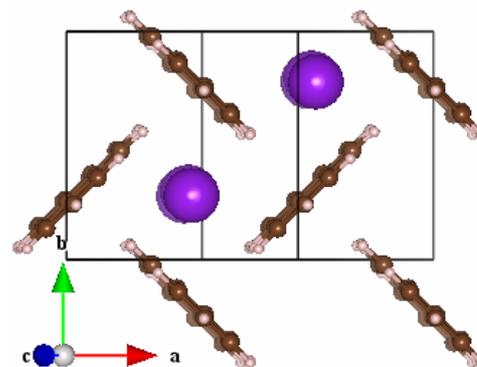
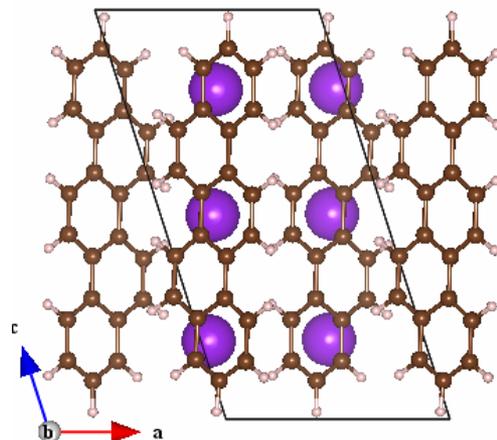
$$b=7.366\text{\AA},$$

$$c=14.209\text{\AA};$$

$$\Omega=106.8^\circ$$

$$V=732.9 \text{\AA}^3$$

GGA-PBE



$$a=7.719\text{\AA},$$

$$b=7.725\text{\AA},$$

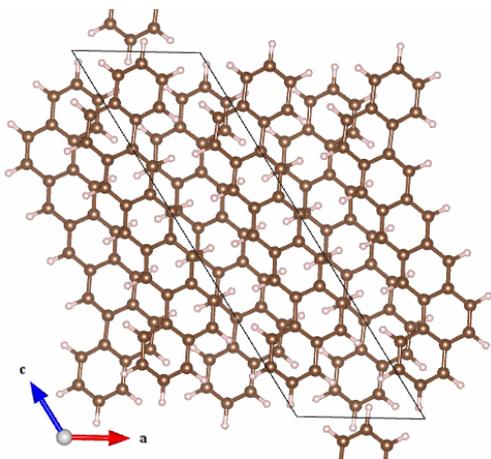
$$c=14.974\text{\AA};$$

$$\Omega=107.6^\circ$$

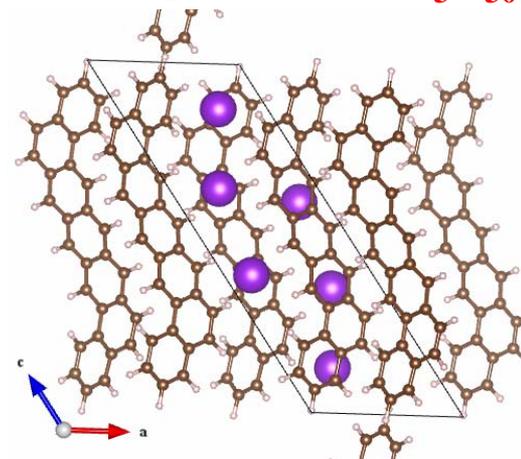
$$V=851.11 \text{\AA}^3$$

$K_3C_{30}H_{18}$ (our results)

1,2:8,9-dibenzopentacene ($C_{30}H_{18}$) K-doped 1,2:8,9-dibenzopentacene ($K_3C_{30}H_{18}$)



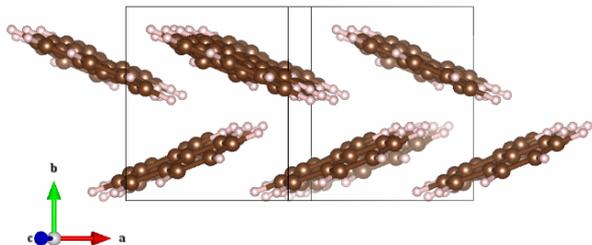
Hypothetical structure



Tri-potassium

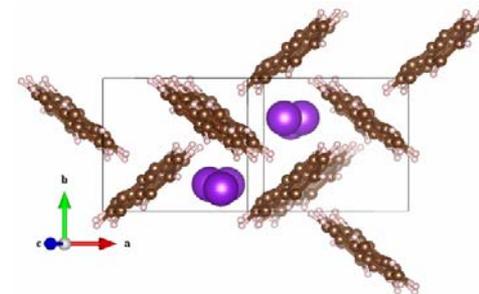


No experimental data



$$V=831.00 \text{ \AA}^3$$

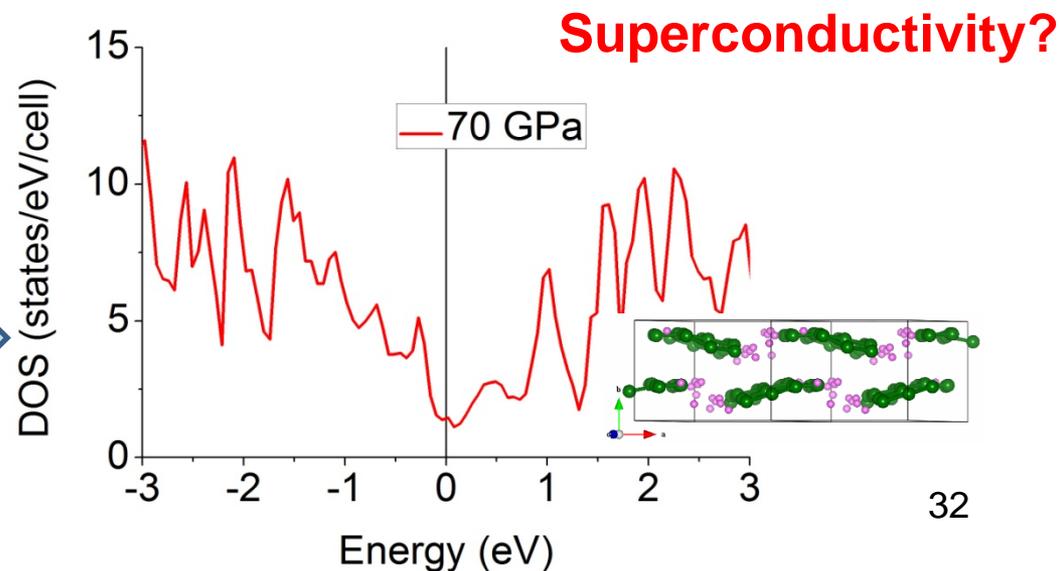
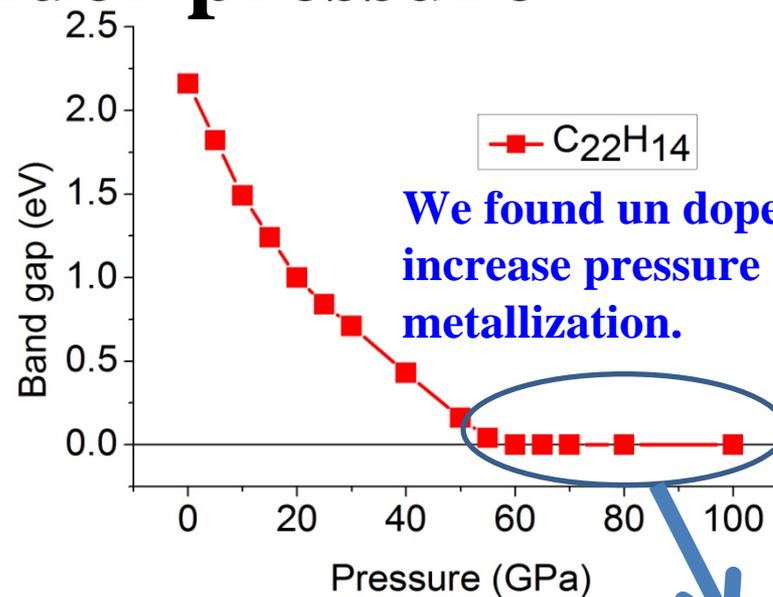
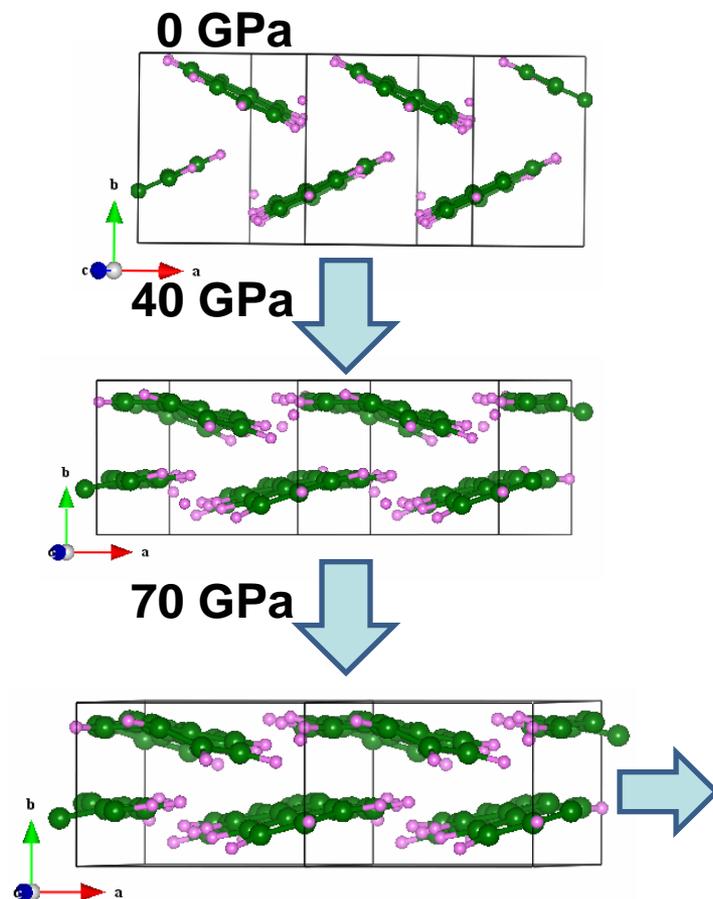
$$a=6.279\text{\AA}, b=7.381\text{\AA}, c=21.225\text{\AA}; \beta =122.35^\circ$$



$$V=962.09 \text{ \AA}^3$$

$$a=7.711\text{\AA}, b=7.041\text{\AA}, c=21.357\text{\AA}; \beta =123.93^\circ$$

$C_{22}H_{14}$ under pressure



Superconductivity of Metallic SiH₄ with Layered Structure

X.J. Chen, J. L. Wang, V.V. Struzhkin, H.K. Mao, R.J. Hemley, and H.Q Lin, (PRL 101, 077002 (2008); Jun 19, 2007)

The electronic and lattice dynamical properties of compressed solid SiH₄ have been calculated in the pressure range up to 300 GPa with density functional theory. We find two energetically preferred insulating phases with $P2_1/c$ and Fdd_2 symmetries at low pressures. We demonstrate that the *Cmca* structure having layered network is the most likely candidate of the metallic phase of SiH₄ over a wide pressure range above **60 GPa**. The superconducting transition temperature in this layered metallic phase is found to be in the range of **20** and **75** K.

Brief Summary

- For undoped hydrocarbon compounds, crystal structures of $C_{14}H_{10}$ and $C_{22}H_{14}$ are experimentally known, but not $C_{30}H_{18}$ and other longer rings.
- For K-doped cases, few structures are known. Searching optimal structures are non-trivial.
- Studies on crystal structure as functions of pressure are called for, especially for doped cases. Usual DFT optimization does not provide results agree well with experiments. Different searching algorithms will find their values here.

Correlation Effect

Electron-Electron Correlation (EC)?

Measure of correlation?

Magnetism?

Previous Results: $C_{22}H_{14}$ (theory)

PHYSICAL REVIEW B 83, 134508 (2011)

Electronic correlation effects in superconducting picene from *ab initio* calculations

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¹ISC-CNR and Dipartimento di Fisica, Università di Roma “La Sapienza,” Piazzale A. Moro 5, I-00185 Rome, Italy

²Institute for Theoretical Solid State Physics, IFW Dresden, D-01171 Dresden, Germany

³Democritos National Simulation Center, CNR-IOM and Scuola Internazionale Superiore di Studi Avanzati (SISSA),

Via Bonomea 265, Trieste, Italy

(Received 16 November 2010; revised manuscript received 21 February 2011; published 7 April 2011)

We show, by means of *ab initio* calculations, that electron-electron correlations play an important role in potassium-doped picene (K_x -picene), recently characterized as a superconductor with $T_c = 18$ K. The inclusion of exchange interactions by means of hybrid functionals reproduces the correct gap for the undoped compound and predicts an antiferromagnetic state for $x = 3$, where superconductivity has been observed. These calculations, which do not require us to assume a value for the interaction strength, indirectly suggest that these materials should have a sizable ratio between the effective Coulomb repulsion U and the bandwidth. This is fully compatible with simple estimates of this ratio. **Using these values of U in a simple effective Hubbard model, an antiferromagnetic state is indeed stabilized.** Our results highlight the similarity between potassium-doped picene and alkali-doped fulleride superconductors.

$$H = \sum_{i\alpha\sigma} \epsilon_i^\alpha c_{i\alpha\sigma}^\dagger c_{i\alpha\sigma} + \sum_{ij\alpha\beta} t^{\alpha\beta} (c_{i\alpha\sigma}^\dagger c_{j\beta\sigma} + \text{H.c.}) + \sum_i \frac{U}{2} n_i^2 \quad 37$$

C₂₂H₁₄ (theory)

PHYSICAL REVIEW B 83, 214510 (2011)

Density functional calculations of electronic structure and magnetic properties of the hydrocarbon K₃picene superconductor near the metal-insulator transition

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We have investigated the electronic structures and magnetic properties of a newly discovered hydrocarbon superconductor, K₃picene, having $T_c = 18$ K. We have shown that the metal-insulator transition is driven in K₃picene by 5% volume enhancement with the formation of the local magnetic moment. Active bands for the superconductivity near the Fermi level (E_F) have hybridized character of LUMO and LUMO + 1 of the picene molecule. Fermi surfaces of K₃picene manifest neither prominent nesting feature nor marked two-dimensional behavior. By estimating the ratio of the Coulomb interaction, U , and the bandwidth, W , of the active bands near E_F , we have demonstrated that K₃picene is located in the vicinity of the Mott transition. Our findings suggest that **K₃picene is a strongly correlated electron system.**

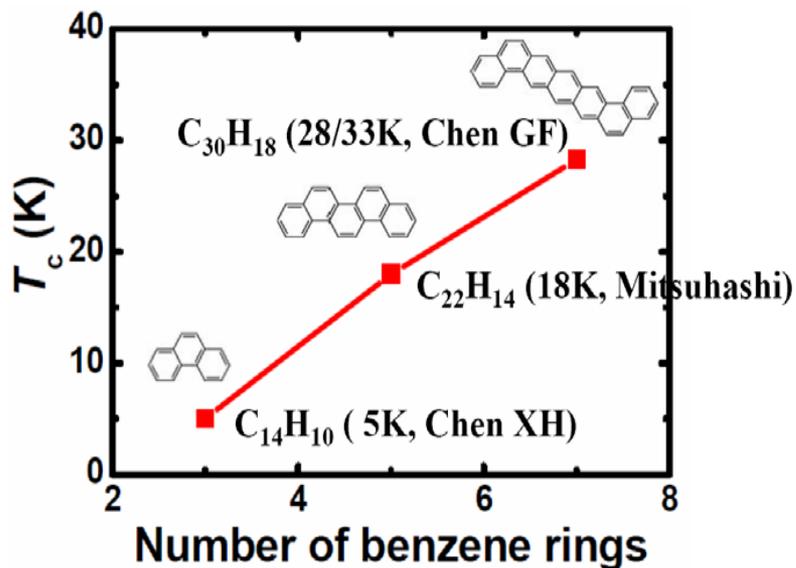
$$U/W = 2.83 > 1.73 \text{ (Mott Insulator); } \text{CuO}_2, U \sim W=8t_{38}$$

$K_3C_{22}H_{14}$ is a strongly correlated electron system.

Then how about $K_3C_{14}H_{10}$ and $K_3C_{30}H_{18}$?

Other similar systems ?

Organic Superconductors



How about the relation between EC and T_c (or structure)?

Our Studies

- Comparisons of electronic correlation effects for **undoped** hydrocarbon compounds
- Comparisons of electronic correlation effects and electronic structures for **doped** hydrocarbon compounds

Method

For undoped systems:

- Package: VASP
- Method: PAW
- Exchange correlation potential:
GGA-PBE, Heyd-Scuseria-Ernzerhof (**HSE**)
- Cutoff energy: 600 eV
- K-point: 6 x 8 x 6 ($C_{14}H_{10}K_3$), 6 x 8 x 6 ($C_{22}H_{14}K_3$), 4 x 4 x 2 ($C_{30}H_{18}K_3$)
- DOS calculations: tetrahedron method

Method

For K-doped systems:

- **Package:** WIEN2k
- **Method:** FP-LAPW
- **Exchange correlation potential:**
GGA-PBE, Heyd-Scuseria-Ernzerhof (**HSE**)
- **Cutoff energy:** -9.0 Ry
- **K-point:** 3 x 4 x 3 ($C_{14}H_{10}K_3$), 3 x 5 x 2 ($C_{22}H_{14}K_3$), 5 x 4 x 1 ($C_{30}H_{18}K_3$)
- **DOS calculations:** tetrahedron method
- **R_{MT} :** 1.29, 0.67, and 2.0 a.u. for C, H, and K, respectively

Estimating Correlation

After getting the lattice structure, we obtain the electronic structures by using the generalized gradient approximation (GGA) [1]. To overcome the limitations of PBE-GGA (and LDA) and its shortcoming in the determination of the gap, we repeated the same kind of calculation by using the hybrid functional method (HSE) [2-4].

- [1] J. P. Perdew, K. Burke, and M. Ernzerhof, Phys. Rev. Lett. **77**, 3865 (1996).
- [2] J. Heyd, G. E. Scuseria, and M. Ernzerhof, J. Chem. Phys. **118**, 8207 (2003).
- [3] J. Heyd and G. E. Scuseria, J. Chem. Phys. **121**, 1187 (2004).
- [4] J. Heyd, G. E. Scuseria, and M. Ernzerhof, J. Chem. Phys. **124**, 219906 (2006).

Based on PBE exchange-correlation functional approach [1], the hybrid density functional, which includes a certain amount of HF exchange, could be written as

$$E_{XC}^{HSE} = \nu E_X^{HF} + (1 - \nu) E_X^{PBE} + E_C^{PBE}$$

Estimating Correlation

$$E_{XC}^{\text{HSE}} = \nu E_X^{\text{HF}} + (1 - \nu) E_X^{\text{PBE}} + E_C^{\text{PBE}}$$

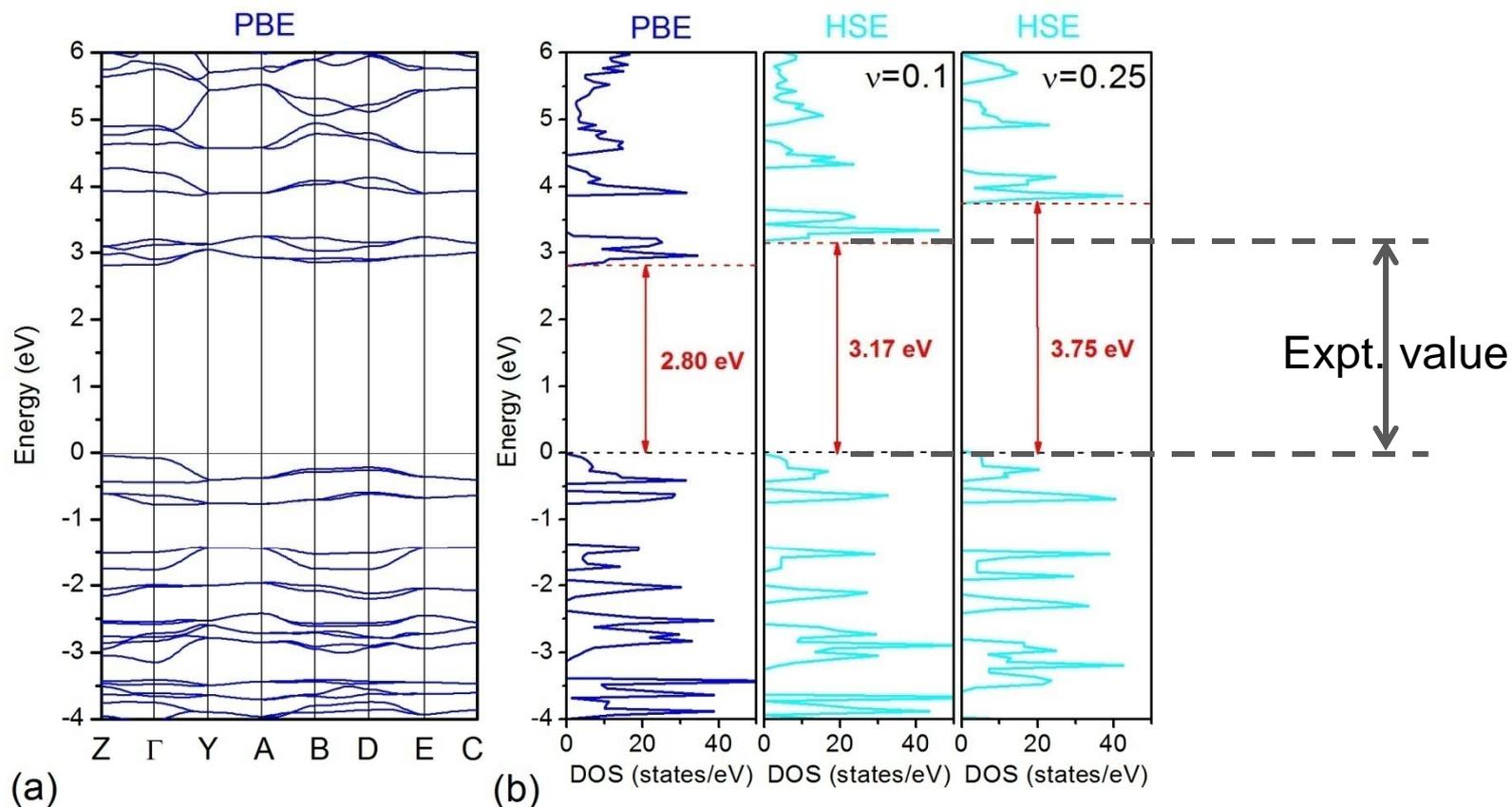
where the E_X^{HF} represents the HF exchange form, the E_X^{PBE} corresponds to the PBE exchange form, and the E_C^{PBE} is the PBE correlation energy.

The mixing coefficient ν represents the amount of the HF exchange interaction and reflects the intensity of electronic correlation (EC).

Introducing a screened Coulomb potential (μ) [20] and splitting all terms into their short-range (SR) and long-range (LR) components:

$$E_{XC}^{\text{HSE}} = \nu E_X^{\text{HF,SR}}(\mu) + (1 - \nu) E_X^{\text{PBE,SR}}(\mu) + E_X^{\text{PBE,LR}}(\mu) + E_C^{\text{PBE}}$$

Coulomb screen μ was adjusted from the 0.1 Å to 0.2 Å. ν was adjusted from 0 to 1, $\nu = 0$ corresponds to PBE.

Pure phenanthrene ($C_{14}H_{10}$):Fixing $\mu = 0.2$ Based on
experimental
lattice
constants

For phenanthrene, the Band-gap obtained from PBE is 2.80 eV, which is less than the experimental value. Small amount ($\nu > 0.1$) of exact exchange (HSE calculation) is added to correct the error. However, normal HSE calculation (adopted default $\nu > 0.25$) will overestimate the gap.

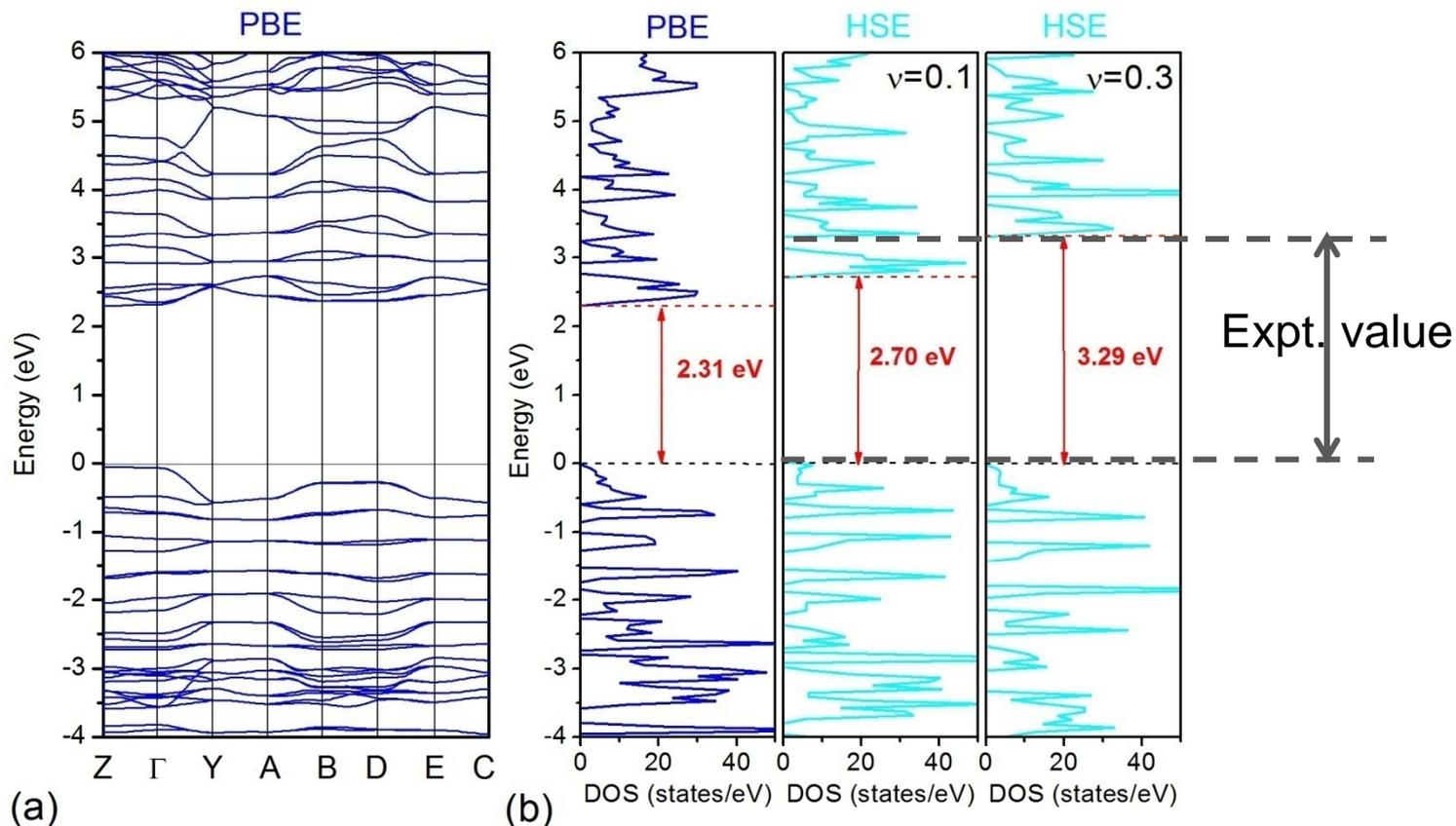
Namely, the electronic correlation effect is not neglected in solid phenanthrene.

Experimental Band-gap: 3.16 eV;

M. T. Bhatti et al., Turk. J. Phys. **24**, 673 (2000).

LDA result: 2.75 eV 47

P. L. de Andres et al. PRB **84**, 144501 (2011)

Pure picene ($C_{22}H_{14}$):Fixing $\mu = 0.2$ Based on
experimental
lattice
constants

For picene, the Band-gap obtained from PBE is only 2.31 eV, the gap is intensively underestimated by PBE. HSE functional induces the increase of gap and furthers with increasing the ratio ν of exact/DFT exchange. When ν is increased to ~ 0.3 , the obtained gap is equal to the experimental value.

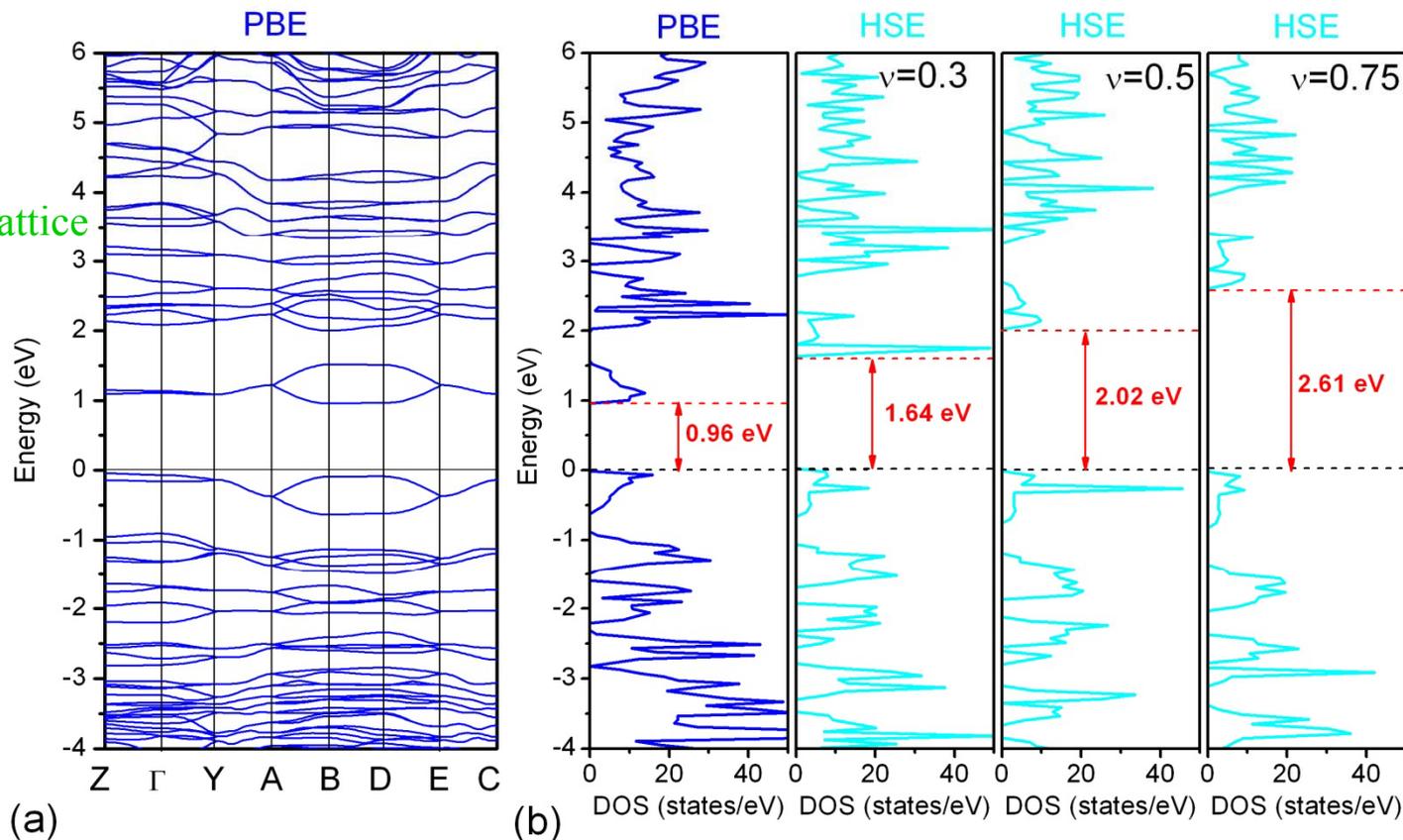
It indicates that the PBE level can not describe the electronic properties of picene truly since the strong electronic correlation effect exists in this system, and more stronger effect than phenanthrene.

Experimental Band-gap: 3.3 eV;

H Okamoto et al., JACS 130, 10470 (2008)

LDA result: 2.36 eV

T. KOSUGI et al., JPSJ 78, 113704 (2009)

Pure 1,2:8,9-dibenzopentacene ($C_{30}H_{18}$):Fixing $\mu = 0.2$ Based on
optimized lattice
constants

For $C_{30}H_{18}$, the Band-gap obtained from PBE is only 0.96 eV, the gap is intensively underestimated by PBE. HSE functional induces the increase of gap and furthers with increasing the ratio ν of exact/DFT exchange. When ν reaches > 0.5 , the obtained gap is consistent with the experimental implication.

It indicates that the PBE level can not describe the electronic properties of $C_{30}H_{18}$ truly since the strong electronic correlation effect exists in this system, and more stronger effect than $C_{14}H_{10}$ and $C_{22}H_{14}$.

Experimental optical band-gap: > 2 eV;

Correlation Measure

Reproducing the experimental band gap, the values of ν and μ parameters are

	$C_{14}H_{10}$	$C_{22}H_{14}$	$C_{30}H_{18}$	$C_{30}H_{18}$
E_g (Expt.)	3.16	3.3	~ 2.0	~ 3.5
ν (fix $\mu=0.1$)	0.05	0.2	~ 0.35	~ 0.75
ν (fix $\mu=0.2$)	0.1	0.3	~ 0.5	> 0.75

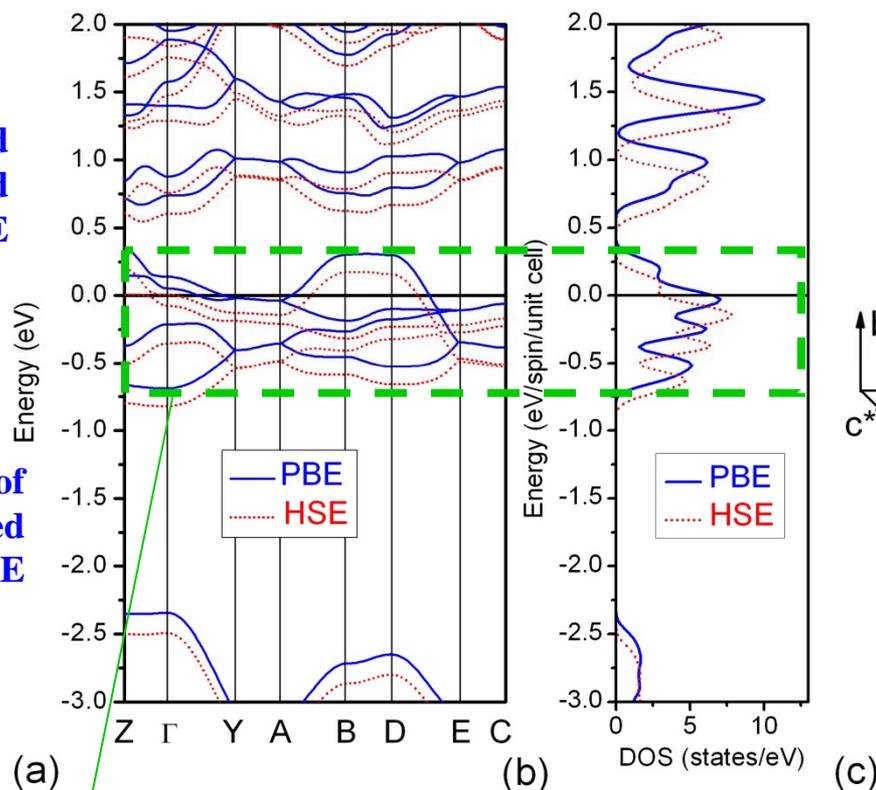
Results show, the ratio of exact/DFT exchange gets larger with the increase of the number of rings, indicating the electron correlations in this system.

For doped cases, considering electronic correlation effects, we suggest the electronic structures of $K_3C_{14}H_{10}$, $K_3C_{22}H_{14}$, and $K_3C_{30}H_{18}$ in the following slides:

Electronic structures of $K_3C_{14}H_{10}$

(a) The band structures based on PBE and HSE levels.

(b) The density of states (DOS) based on PBE and HSE levels.



(c) The Fermi surface sheets based on PBE and HSE levels.

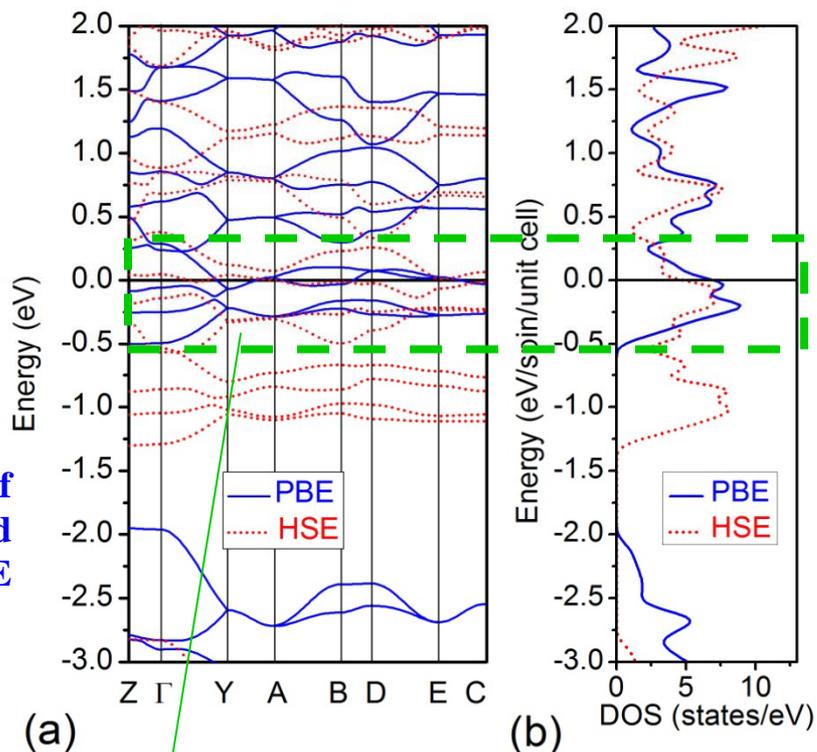
Based on the HSE method, bands shift toward to the lower energy level, results in the decrease of the DOS at Fermi level and variations of Fermi surface. **No clear magnetic ground state, metallic behavior.**

The wide band dispersion implies the **weak correlation** in $K_3C_{14}H_{10}$.

Electronic structures of $K_3C_{22}H_{14}$

(a) The band structures based on PBE and HSE levels.

(b) The density of states (DOS) based on PBE and HSE levels.



(c) The Fermi surface sheets based on PBE and HSE levels.

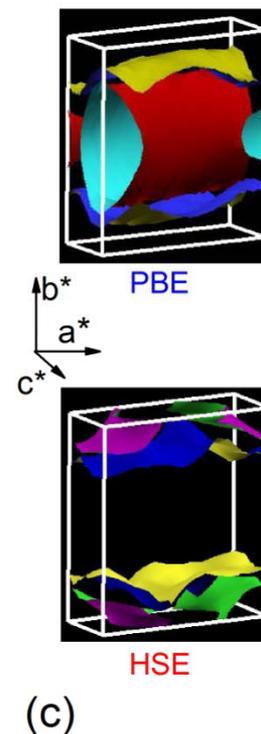
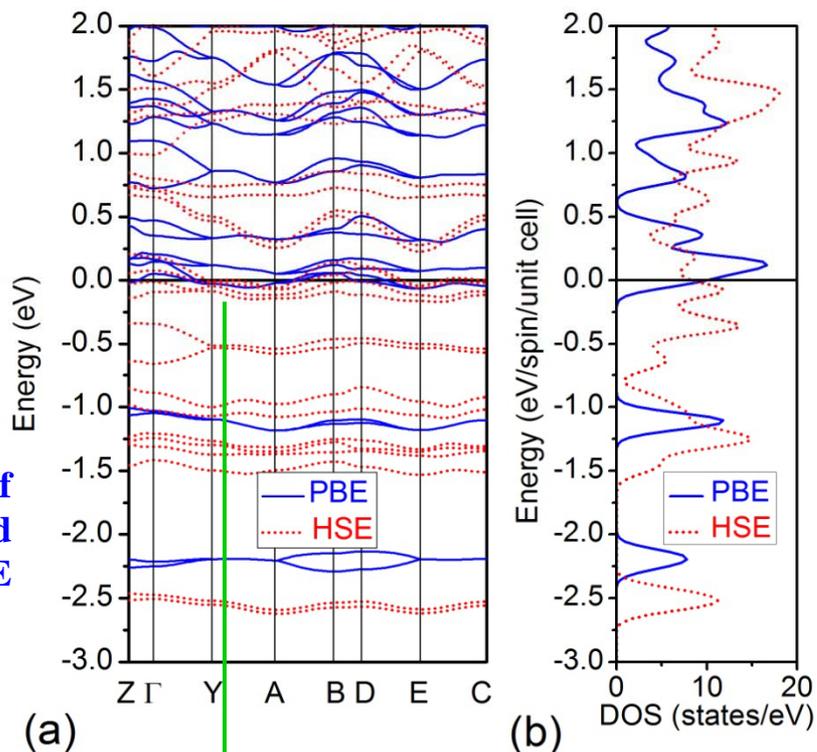
The HSE method results in the decrease of the DOS at Fermi level and variations of Fermi surface.

The feature of **flat bands** indicates that the correlation effect in $K_3C_{22}H_{14}$ is stronger than that in $K_3C_{14}H_{10}$.

Electronic structures of $K_3C_{30}H_{18}$

(a) The band structures based on PBE and HSE levels.

(b) The density of states (DOS) based on PBE and HSE levels.



(c) The Fermi surface sheets based on PBE and HSE levels.

The HSE method results in the big variation of bands and DOS near Fermi level. **Metal, trends to AFM ground-state.**

In $K_3C_{30}H_{18}$, bands become **narrow and more flat**, showing that the **correlation is more stronger**, comparing with $K_3C_{22}H_{14}$.

Magnetism

- Un doped $C_{14}H_{10}$, $C_{22}H_{14}$, and $C_{30}H_{18}$ are non-magnetic.
- Doped $K_3 - C_{14}H_{10}$, $C_{22}H_{14}$, and $C_{30}H_{18}$:

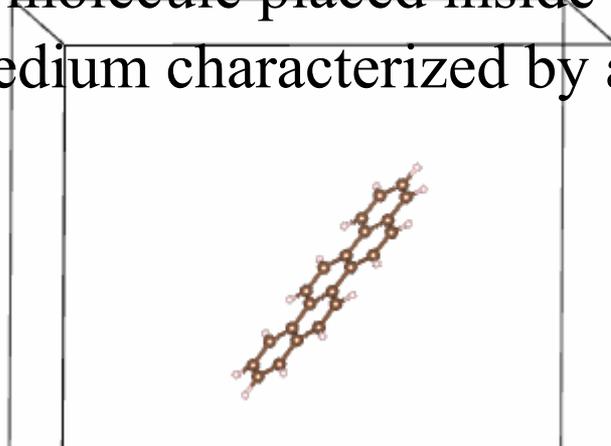
Relative energies to magnetic ground state (unit: meV) based on the experimental lattice constants. AFM-1 represents the initial spin antiparalleling in molecular plane formed by three (or five) rings, while AFM-2 marks the initial spin antiparalleling between two molecular layers and paralleling in molecular plane.

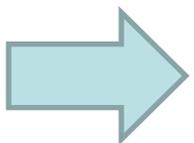
	functional	NM	FM	AFM-1	AFM-2	<i>M</i>
$K_3C_{14}H_{10}$	PBE	0.5	1.0	7.5	0	0.09
	HSE	2.0	1.4	7.7	0	0.13
$K_3C_{22}H_{14}$	PBE	6.2	6.2	6.9	0	0.25
	HSE	7.9	7.9	10.3	0	0.30
$K_3C_{30}H_{18}$	PBE	30.6	20.3	57.0	0	0.45
	HSE	97.2	80.8	103.4	0	0.49

For $K_3C_{14}H_{10}$, no visible magnetic ground-state, while $K_3C_{30}H_{18}$ seems to have AFM-2 state. Correlation increases with number of benzene rings.

$$U_{\text{eff}}, W, t = U_{\text{eff}}/W$$

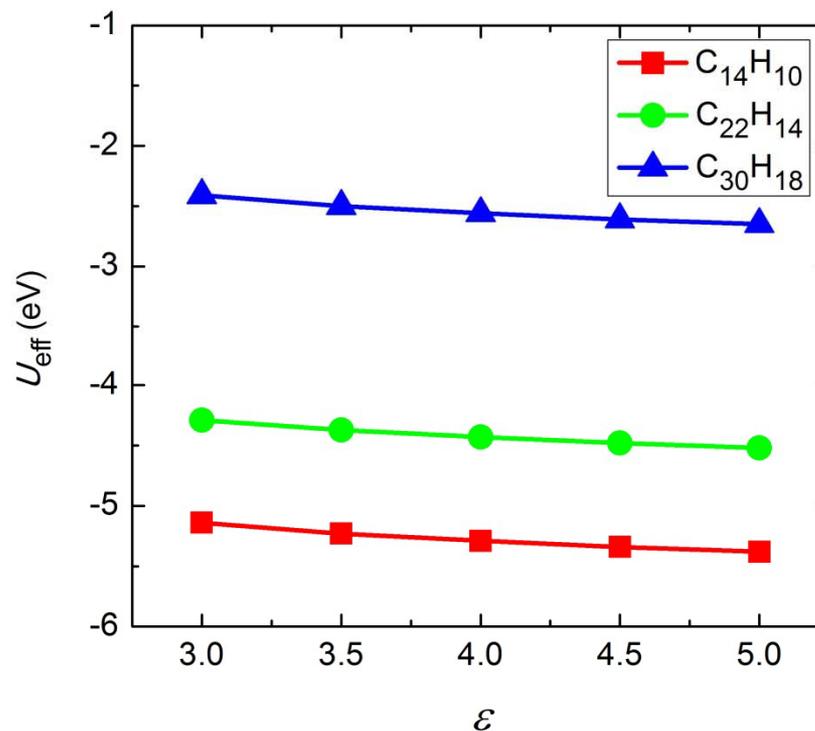
- We compute U_{bare} for a single molecule with three electrons from the energies of the molecule charged with 2, 3, and 4 electrons as $U_{\text{bare}} = E(4) - 2E(3) + E(2)$, where $E(M)$ is the total energy of a molecule charged with M extra electrons.
- Such an estimate for an isolated molecule needs to be corrected ($E_{\text{pol}}(M)$) in order to include the screening effects in the solid. A first estimate can be obtained by considering the effect of the polarization of a charged molecule placed inside a cavity of a homogeneous dielectric medium characterized by a dielectric constant ϵ .
- Thus, we estimate, $U_{\text{eff}} = U_{\text{bare}} - E_{\text{pol}}$





U_{eff} for undoped systems

	ϵ	U_{eff} (eV)
$\text{C}_{14}\text{H}_{10}$	3.0	-5.14
	3.5	-5.23
	4.0	-5.29
	4.5	-5.34
	5.0	-5.38
$\text{C}_{22}\text{H}_{14}$	3.0	-4.29
	3.5	-4.37
	4.0	-4.43
	4.5	-4.48
	5.0	-4.52
$\text{C}_{30}\text{H}_{18}$	3.0	-2.41
	3.5	-2.50
	4.0	-2.56
	4.5	-2.61
	5.0	-2.65

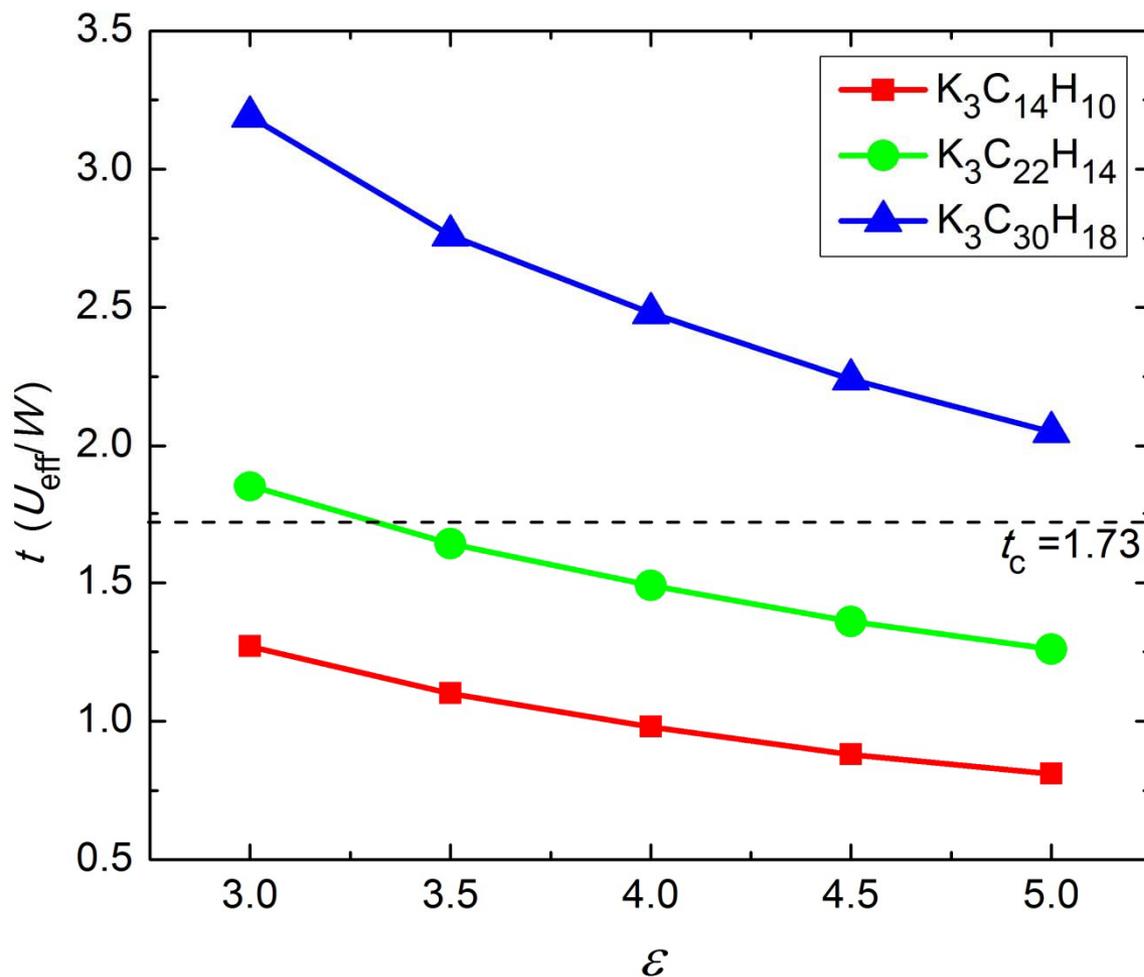


For each hydrocarbon compound, the U_{eff} dependence on dielectric constant ϵ .

$$U_{\text{eff}}, W, t = U_{\text{eff}}/W$$

	ϵ	U_{eff} (eV)	W (eV)	$t = U_{\text{eff}}/W$
$\text{K}_3\text{C}_{14}\text{H}_{10}$	3.0	0.66	0.52	1.27
	3.5	0.57	0.52	1.10
	4.0	0.51	0.52	0.98
	4.5	0.46	0.52	0.88
	5.0	0.42	0.52	0.81
$\text{K}_3\text{C}_{22}\text{H}_{14}$	3.0	0.72	0.39	1.85
	3.5	0.64	0.39	1.64
	4.0	0.58	0.39	1.49
	4.5	0.53	0.39	1.36
	5.0	0.49	0.39	1.26
$\text{K}_3\text{C}_{30}\text{H}_{18}$	3.0	0.67	0.21	3.19
	3.5	0.58	0.21	2.76
	4.0	0.52	0.21	2.48
	4.5	0.47	0.21	2.24
	5.0	0.43	0.21	2.05

$$U_{\text{eff}}, W, t = U_{\text{eff}}/W$$



The $t (U_{\text{eff}}/W)$ dependence on dielectric constant ϵ .

Brief Summary

- For undoped hydrocarbon compounds, the electronic correlation effect (EC) increases from $C_{14}H_{10}$ to $C_{22}H_{14}$ to $C_{30}H_{18}$.
- For K-doped cases, the flat bands near the Fermi level indicate that EC also follows the same pattern.
- Although the existence of the long-rang magnetic order is still interrogative, the appearance of local magnetic moment and its increase with the number of benzene rings establish that the EC increases with the number of benzene rings in the K-doped systems.

Magnetism and Superconductivity

Model?

Magnetism?

Superconductivity?

C₂₂H₁₄ (experiment)

PHYSICAL REVIEW B 82, 195114 (2010)

Electronic structure of pristine and K-doped solid picene: Nonrigid band change and its implication for electron-intramolecular-vibration interaction

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(Received 27 August 2010; revised manuscript received 14 October 2010; published 10 November 2010)

We use **photoemission spectroscopy** to study electronic structures of pristine and K-doped solid picene. The valence band spectrum of pristine picene consists of three main features with no state at the Fermi level (E_F) while that of K-doped picene has three structures similar to those of pristine picene with new states near E_F , consistent with the semiconductor-metal transition. The K-induced change cannot be explained with a simple rigid-band model of pristine picene but can be interpreted by molecular-orbital calculations considering electron-intramolecular-vibration interaction. Excellent agreement of the K-doped spectrum with the calculations points to importance of electron-intramolecular-vibration interaction in K-doped picene.

Electron-phonon interaction exists

$C_{22}H_{14}$ (theory)

PHYSICAL REVIEW B 84, 020508(R) (2011)

Vibrational spectrum and electron-phonon coupling of doped solid picene from first principles

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(Received 21 March 2011; revised manuscript received 5 July 2011; published 25 July 2011)

We study superconductivity in doped solid picene ($C_{22}H_{14}$) with linear response calculations of the phonon spectrum and electron-phonon (ep) interaction. We show that the coupling of the high-energy C bond-stretching phonons to the π molecular orbitals for a doping of ~ 3 electrons per picene molecule is sufficiently strong to reproduce the experimental T_c of 18 K within Migdal-Eliashberg theory. For hole doping, we predict a similar coupling leading to a maximum T_c of 6 K. However, we argue that, due to its molecular nature, **picene may belong to the same class of strongly correlated ep superconductors as fullerenes**. We propose several experimental tests for this hypothesis and suggest that intercalated hydrocarbons with different arrangements and numbers of benzene rings may be used to study the interplay between ep interaction and strong electronic correlations in the highly nonadiabatic limit.

Found a large coupling of bond-stretching phonons at 1400 and 1600 cm^{-1} both to electrons in the four lowest conduction bands and holes in the two highest valence bands, ..., sufficiently strong to explain $T_c=18K$ in Migdal-Eliashberg theory.

C₂₂H₁₄ (theory)

PRL **107**, 077001 (2011)

PHYSICAL REVIEW LETTERS

week ending
12 AUGUST 2011

Strong Intramolecular Electron-Phonon Coupling in the Negatively Charged Aromatic Superconductor Picene

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(Received 7 April 2011; published 9 August 2011)

Superconductivity was recently discovered in solid potassium-intercalated picene (K₃22ph), in which the picene molecule becomes trianionic (22ph³⁻). In this Letter, we conduct a theory-based study of the superconductivity of 22ph³⁻ within the framework of BCS theory. We estimate the density of states $N(\epsilon_F)$ on the Fermi level to be 2.2 states per (eV molecule spin) by using the theoretical intramolecular electron-phonon coupling l_x and the experimental superconducting transition temperature T_c of 18 K. The theoretical value is consistent with the 1.2 states per (eV molecule spin) determined experimentally for K₃22ph with $T_c = 18$ K, indicating the validity of our theoretical treatment and the electron-phonon mechanism for superconductivity. The predicted l_x , 0.206 eV, for 22ph³⁻ is larger than any value reported for organic superconductors, so picene may have the largest l_x among the superconductors reported so far.

Electron-phonon interaction is important

C₂₂H₁₄ (theory)

PRL 107, 137006 (2011)

PHYSICAL REVIEW LETTERS

week ending
23 SEPTEMBER 2011

Intercalant and Intermolecular Phonon Assisted Superconductivity in K-Doped Picene

Michele Casula,¹ Matteo Calandra,¹ Gianni Profeta,² and Francesco Mauri¹¹*CNRS and Institut de Minéralogie et de Physique des Milieux condensés,
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and Max-Planck Institute of Microstructure Physics, Weinberg 2, 06120 Halle, Germany*

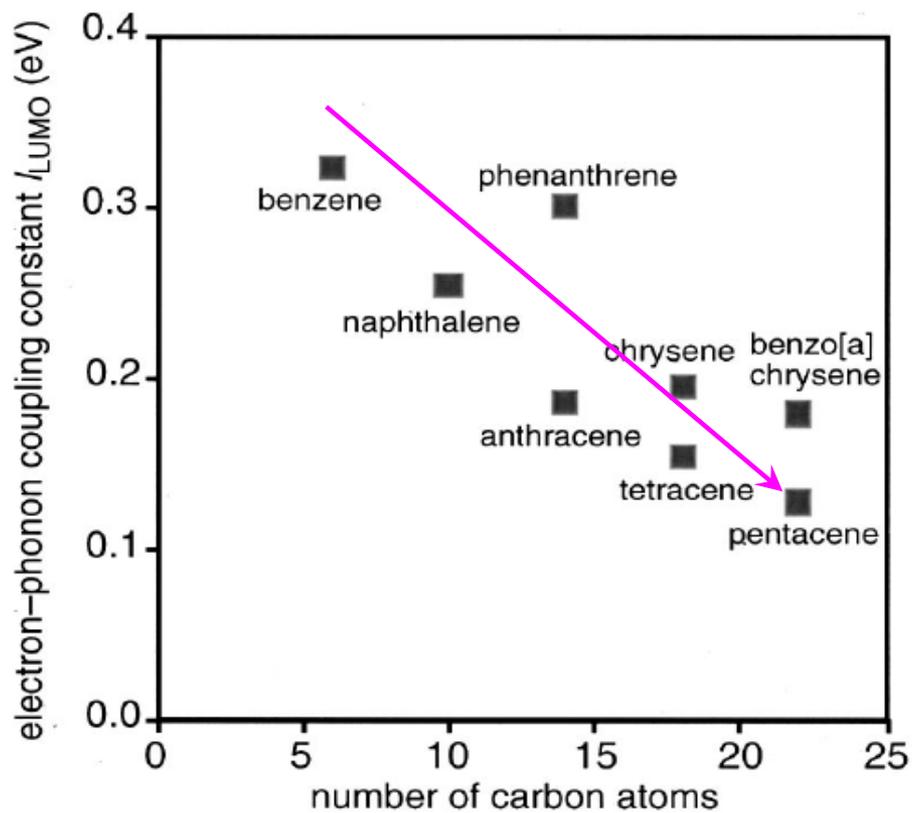
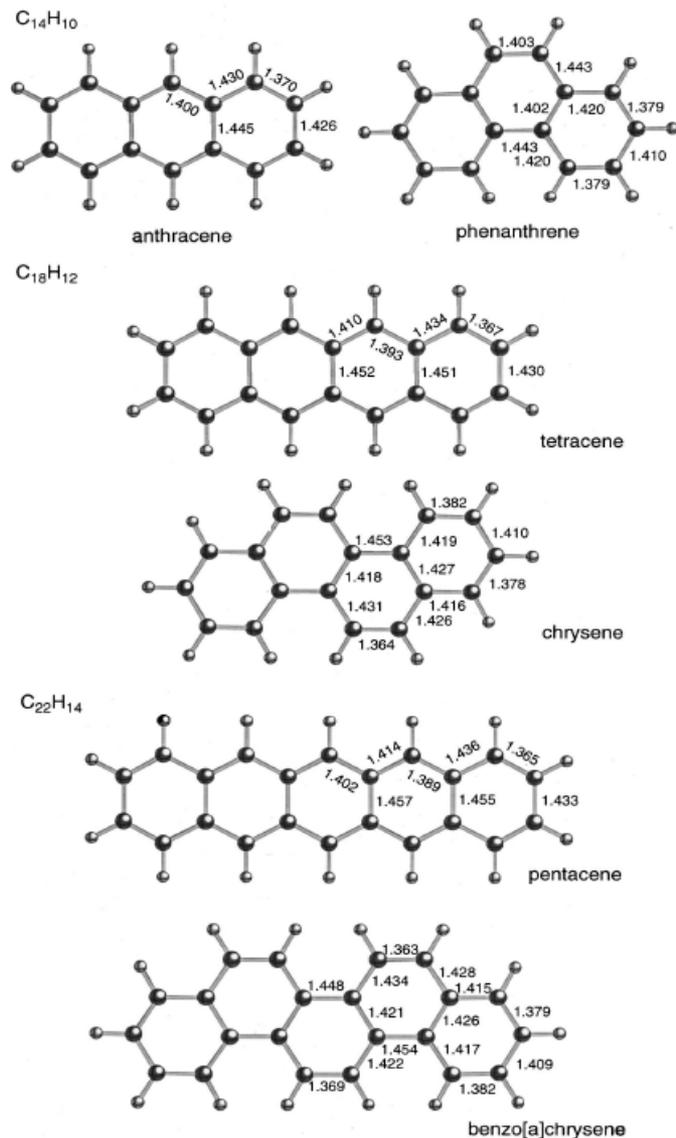
(Received 8 June 2011; published 21 September 2011)

$$\lambda = 0.73, \quad \omega_{\log} = 18.0 \text{ meV (208 K)}$$

Intercalant and intermolecular phonon modes contribute substantially (40%) to λ as also shown by the **isotope exponents** of potassium (0.19) and carbon (0.31).

Other calculations: $\lambda = 0.88/1.41$, $\omega_{\log} = 207/240 \text{ cm}^{-1}$
(Note that intramolecular phonons dominate in C₆₀) 73

Electron-Phonon alone is not enough



T. Kato et al., *J. Chem. Phys.* 116, 3420 (2002). 74

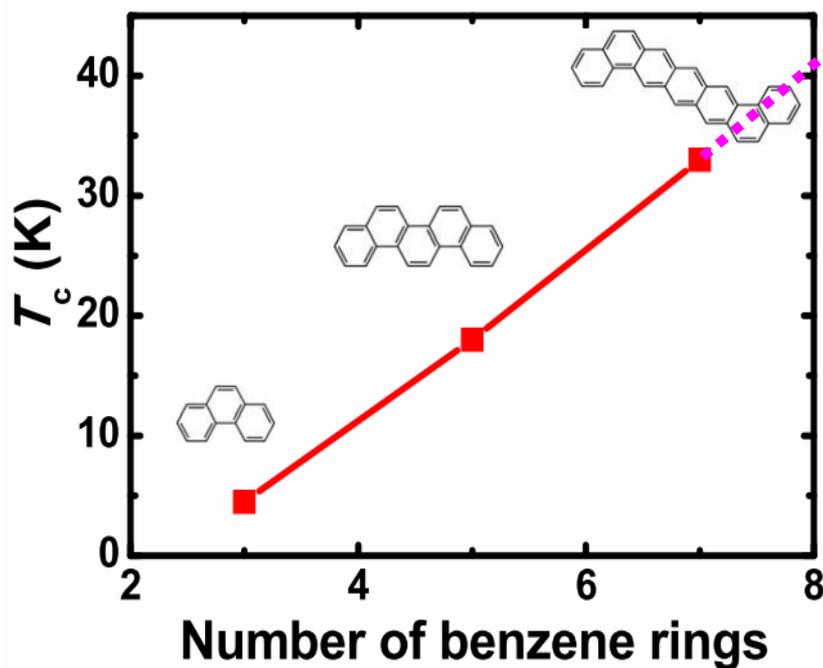
Brief Summary

- Electron-phonon interaction does play role;
- We have seen that electron-electron correlation increases with the number of benzene rings;
- Electron-electron interaction is larger than the bandwidth near the Fermi energy, indicating that correlation should be taken into account.

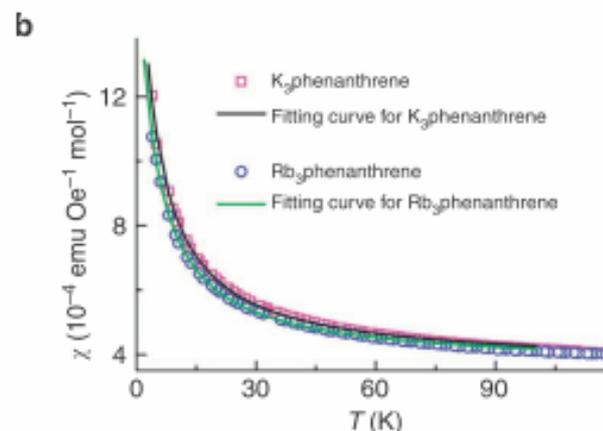
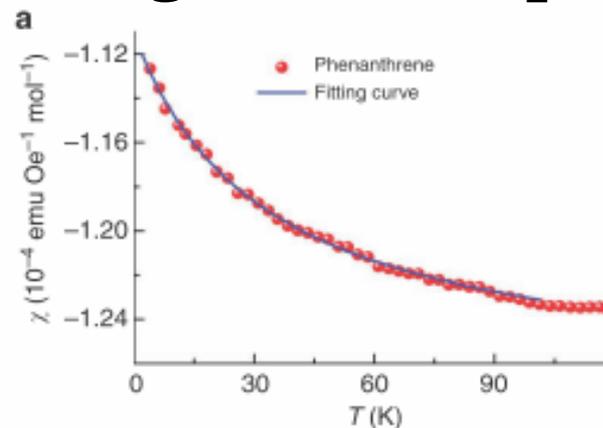
Issues to Address

High magnetic susceptibility

SCIENTIFIC REPORTS | 2 : 389 | DOI: 10.1038/srep00389



T_c increases with increasing the Number of benzene rings!



Curie-Weiss magnetic behavior
→ local spin!

Magnetic Properties

nature

Vol 464 | 4 March 2010 | doi:10.1038/nature08859

LETTERS

Superconductivity in alkali-metal-doped picene

Table 1 | List of A_x picene (A: alkali-metal) samples prepared in this study

A	x	Annealing temperature (K)	Annealing time (days)	Physical properties	Shielding fraction
K	1.0	440	6.5	Pauli-like	NA
K	1.8	440	7.0	Pauli-like	NA
K	2.6	440	8.0	SC ($T_c = 6.5$ K)	$\ll 0.1\%$
K[†]	2.9	440	9.0	SC ($T_c = 7.0$ K)	0.1%
K	3.0	440	8.0	SC ($T_c = 6.5$ K)	$\ll 0.1\%$
K	3.0	440	9.0	SC ($T_c = 17$ K)	0.1%
K	3.1	440	4.0	SC ($T_c = 7.4$ K)	$< 0.1\%$
K	3.3	440	21.0	SC ($T_c = 8$ K)	$\ll 0.1\%$
*K[‡]	3.3	440	21.0	SC ($T_c = 6.9$ K)	15%
K	3.3	440	8.5	SC ($T_c = 7.1$ K)	$\ll 0.1\%$
K	3.3	440	11.0	SC ($T_c = 18$ K)	0.55%
*K[§]	3.3	440	11.0	SC ($T_c = 18$ K)	1.2%
K	4.0	440	8.0	Curie-like	NA
K	5.1	440	12.5	Curie-like	NA
Na	3.4	570	5.0	Pauli-like	NA
Rb	2.8	440	16.5	Pauli-like	NA
RbII	3.1	570	6.7	SC ($T_c = 6.9$ K)	10%
Cs	3.0	440	9.0	Metal-insulator transition	NA

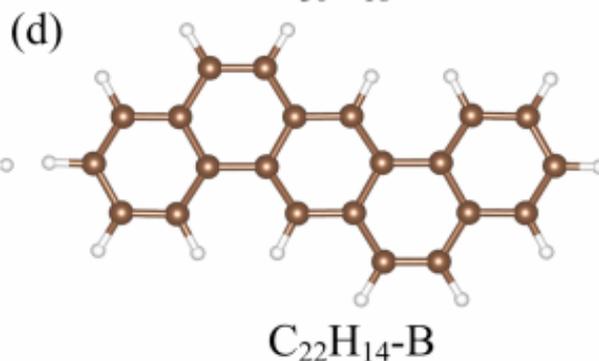
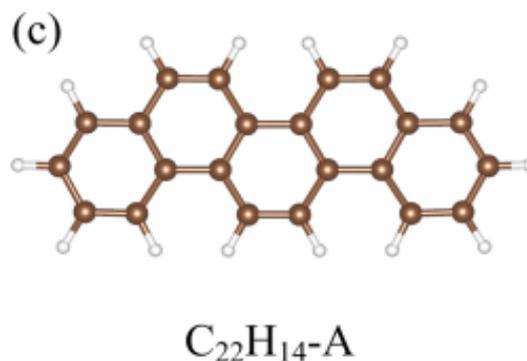
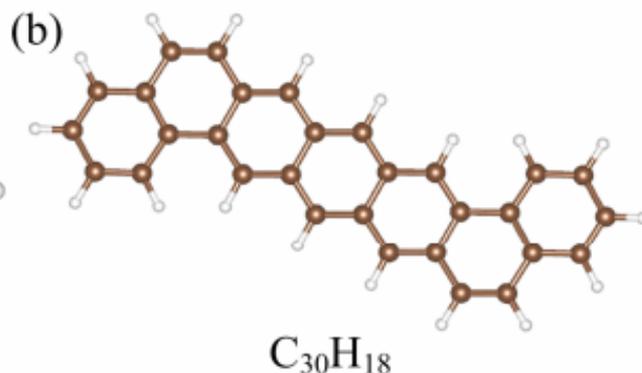
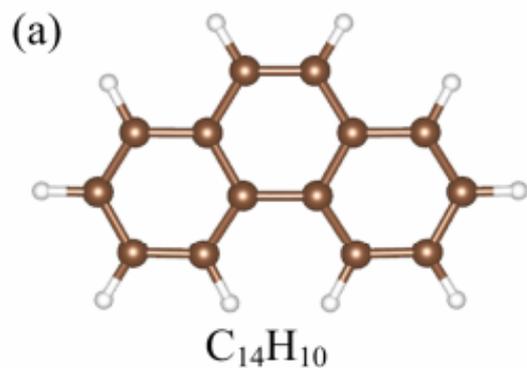
**Normal state:
Curie-like**

ED/QMC Calculation

- Try the simplest model to address the following issues: (1) origin of the local spin? (2) electron correlation contributes to superconductivity?
- Non-perturbative studies on simple systems
- Results:
 - magnetic instability
 - electron pairing instability

Starting point: a Hubbard-like Model

$$H = -t \sum_{\langle ij \rangle, \sigma} (c_{i\sigma}^\dagger c_{j\sigma} + c_{j\sigma}^\dagger c_{i\sigma}) - \epsilon \sum_i n_i + U \sum_i n_{i\uparrow} n_{i\downarrow}$$



$t \sim 2.7-3.0 \text{ eV}$

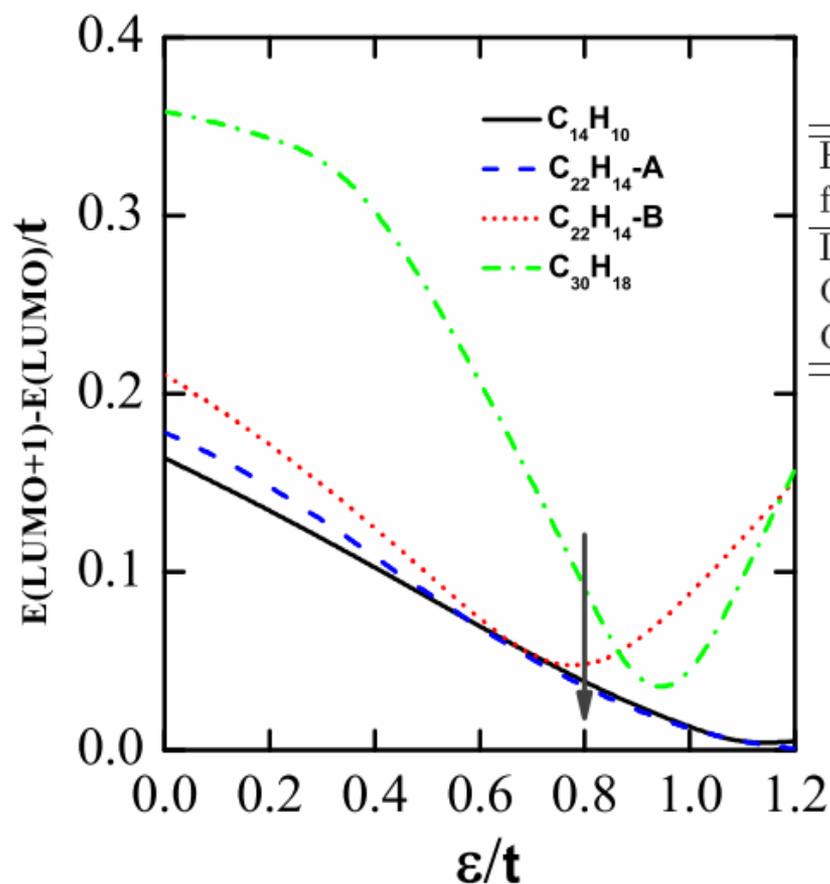
$U \sim 6-10 \text{ eV}$

$U/t \sim 2-3$

$\epsilon ? (\sim t)$

**Electron-phonon
interaction could
be accounted by
 $t(1 + \alpha x)$**

The Parameter ϵ



$$\Delta E_{LUMO} = E(LUMO + 1) - E(LUMO)$$

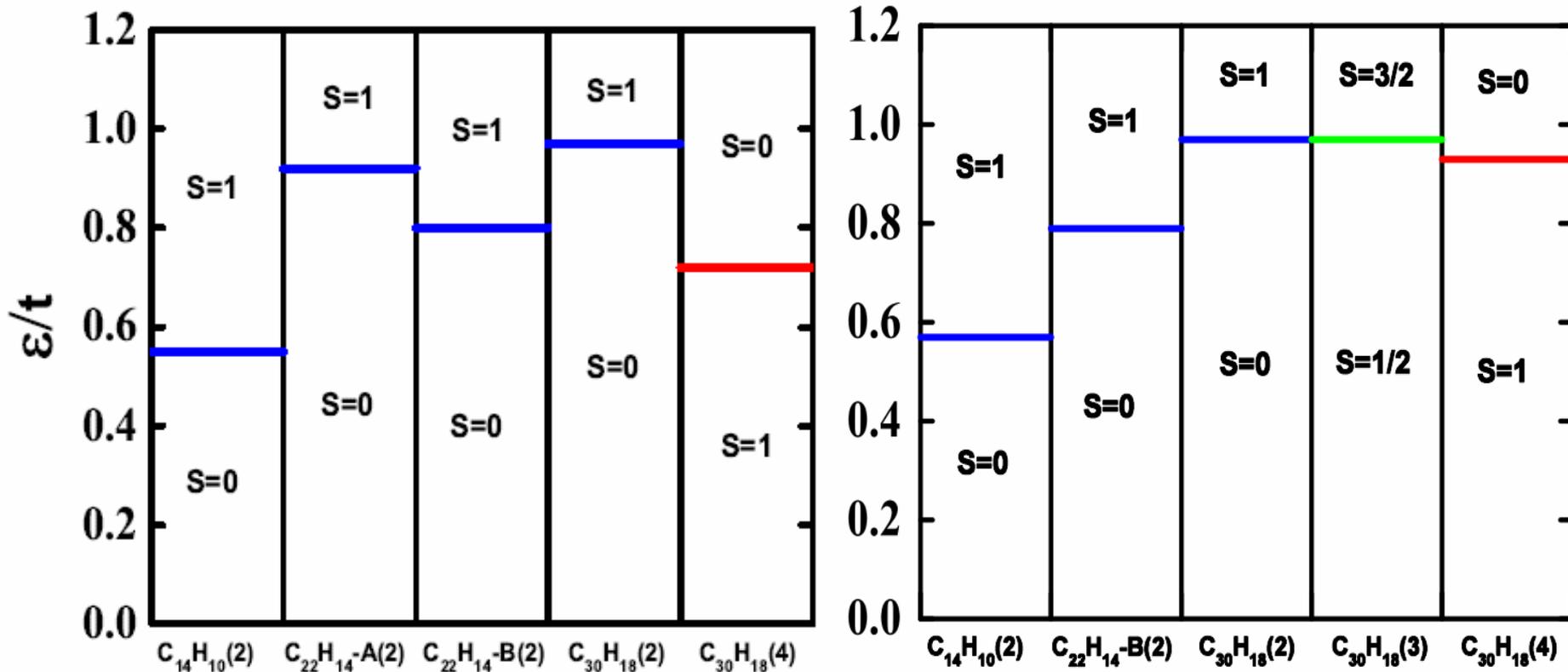
Exchange-correlation function	$\Delta E_{LUMO}(C_{14}H_{10})$	$\Delta E_{LUMO}(C_{22}H_{14})$
LDA	0.1018 eV	0.0634 eV
GGA-PBE	0.1141 eV	0.0750 eV
GGA-PW91	0.1117 eV	0.0750 eV

↓
 $\epsilon > 0.8t$

Magnetic Phase Diagram

$U=2t$

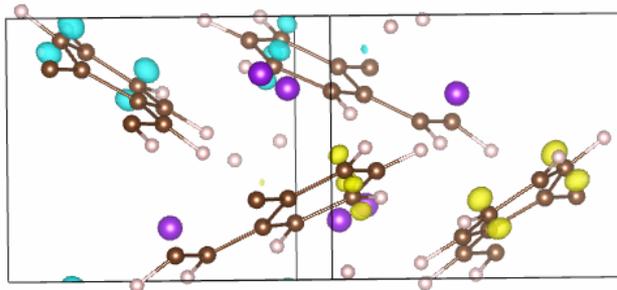
$U=3t$



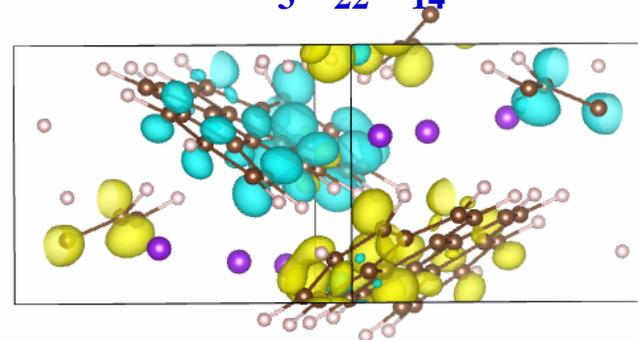
- (1) For two electron doped cases, the total spin S switches from 0 to 1.
- (2) For $C_{30}H_{18}$ with four electrons doped, there is $S=1$ to 0 transition; with three electrons doped, $S=1/2$ to $3/2$.

Spin Density

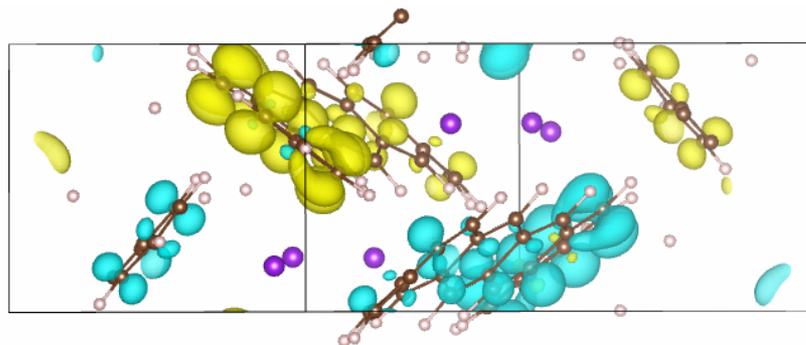
(PBE results)



Local magnetic moment $\sim 0.09 \mu_{\text{B}}/\text{f.u.}$



Local magnetic moment $\sim 0.25 \mu_{\text{B}}/\text{f.u.}$



Local magnetic moment $\sim 0.45 \mu_{\text{B}}/\text{f.u.}$

Effect of nearest-neighbor Coulomb interaction V on the ground state.

TABLE I: V dependence of the energy difference $\Delta E = E(S = 1) - E(S = 0)$ for the charged molecules with two added electrons at $U = 2t$ and $\epsilon = 1.0t$. Statistical errors are in the last digit and shown in the parentheses.

V	$\Delta E(C_{14}H_{10})$	$\Delta E(C_{22}H_{14} - A)$	$\Delta E(C_{22}H_{14} - B)$
0.0	-0.0452(6)	-0.0074(8)	-0.1073(7)
0.1	-0.0305(6)	0.0009(8)	-0.1041(9)
0.2	-0.0016(8)	0.0121(9)	-0.1017(9)
0.3	-0.0044(8)	0.0257(9)	-0.0965(9)
0.4	0.0134(9)	0.043(1)	-0.090(1)
0.5	0.0289(9)	0.060(1)	-0.086(1)

The energy difference increases with increasing V , making $S=0$ state more stable than $S=1$ for large V .

Pair Binding Energy:

$$\Delta_i = 2(E_i - E_{i-1}) - (E_{i+1} - E_{i-1}) = 2E_i - E_{i-1} - E_{i+1}$$

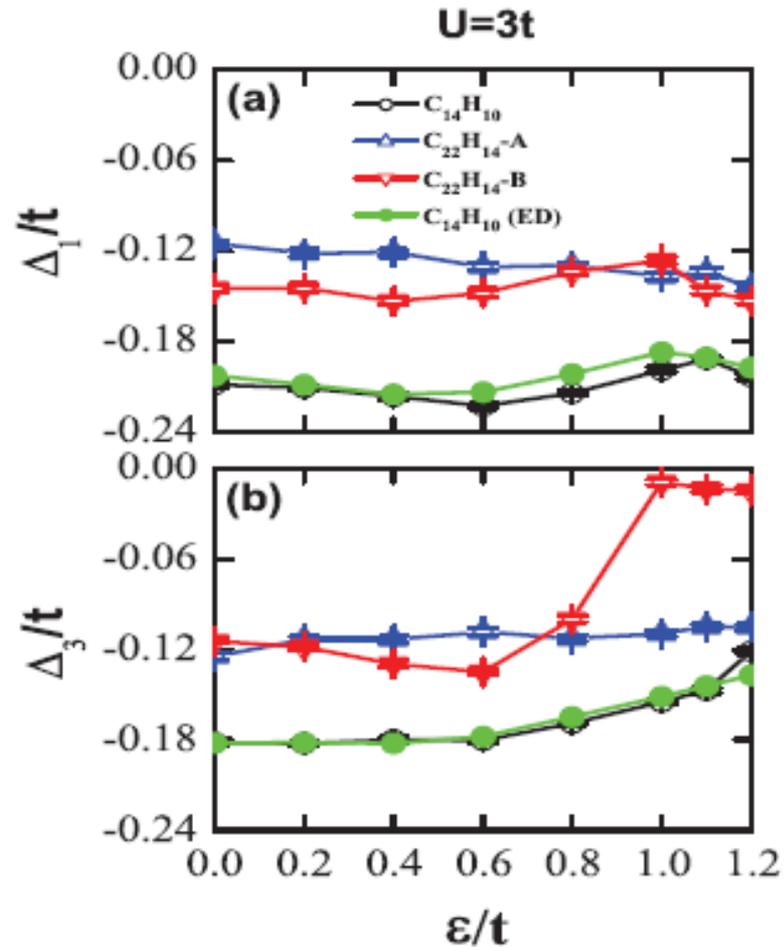
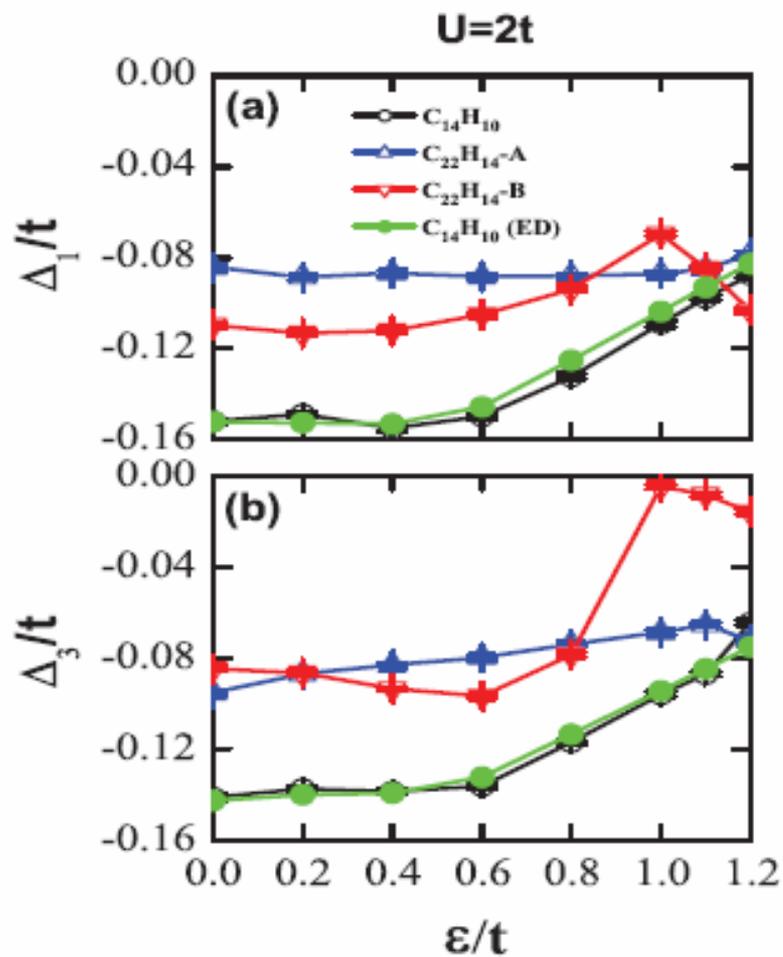
$i=1, 2, 3$

$\Delta_i > 0$: attractive interaction for added electrons

$\Delta_i < 0$: repulsive interaction for added electrons

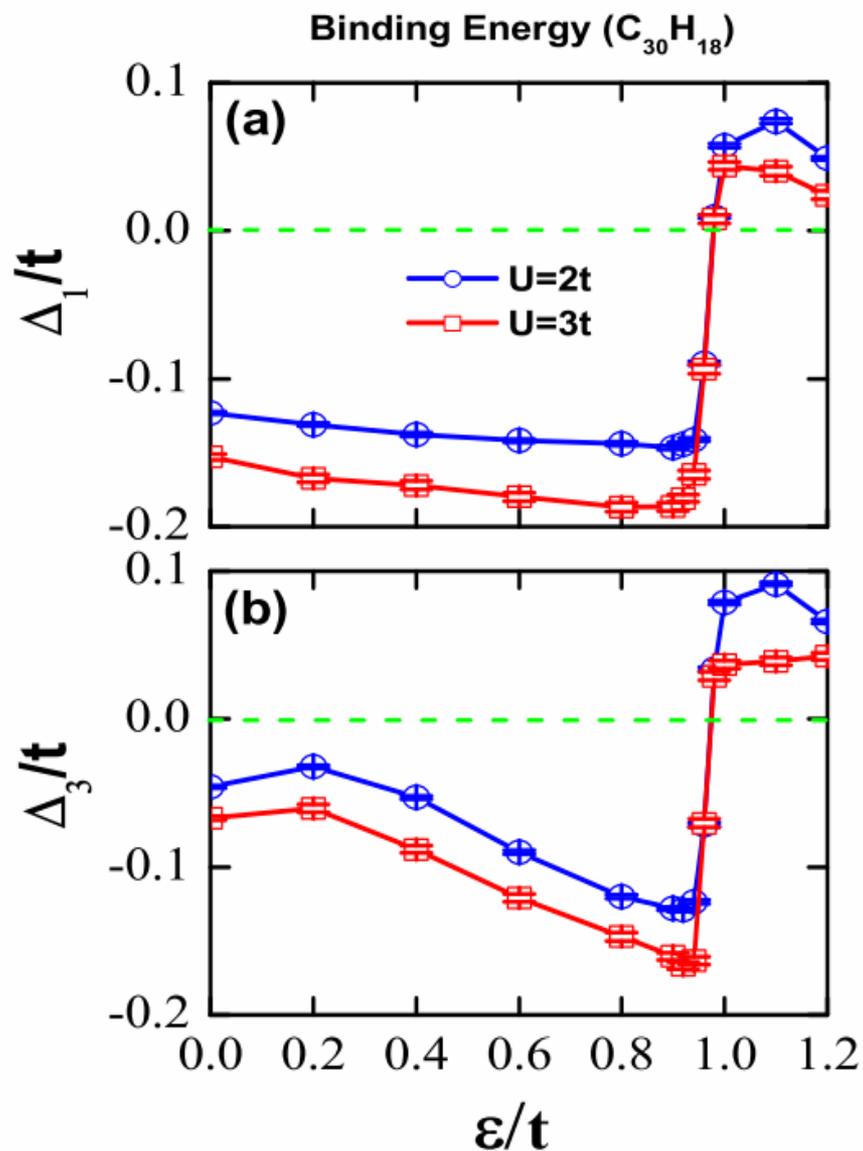
- (1) Applied for the low energy physics of renormalized electrons, not for the bare electrons.
- (2) Applied for high-T_c superconductors and C₆₀.

Pair Binding Energy Δ_i for $C_{14}H_{10}$ & $C_{22}H_{14}$



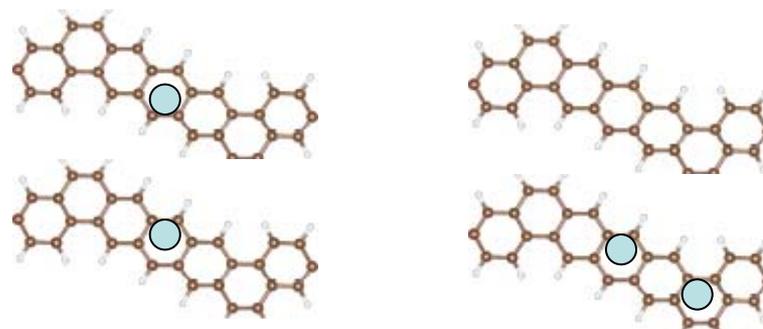
Δ_i is always negative \Rightarrow hardly to form Cooper pairs.

Pair Binding Energy Δ_i for $C_{30}H_{18}$



• Pair binding energy is positive for $\epsilon > 1.0t$.

• Electron attraction through charge disproportionation:



$E_L > E_R$

Brief Summary

- The spin polarized state is realized in the charge aromatic molecules of PAHs, especially for even number of added electrons.
- Electron correlations might contribute to superconductivity. Larger molecules with a similar structure to $C_{30}H_{18}$ may exhibit higher superconducting transition temperature.
- Parameters are consistent with DFT results.

Summary and Discussions

- We searched for possible structures of Polycyclic Aromatic Hydrocarbon. More systematic works are needed, including pressure effects.
- We found that there exists strong correlation and it increases with the size of benzene rings.
- We proposed a model to address magnetic and superconducting properties. Our results showed that there exist local moment and pairing instability. Both of them increase with the size of benzene rings. Such behavior is very different from other systems.



Thank You !



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PERSPECTIVE

Metal-intercalated aromatic hydrocarbons: a new class of carbon-based superconductors

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